



Trends in Nitrogen Deposition to the Chesapeake Bay Watershed 2002-2012

Jesse Bash, Robin Dennis, Ellen Cooter and Others

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Outline

- Review of advancements in the CMAQ modeling system
- Trends in N deposition to Chesapeake Bay Watershed
- What is causing these changes?
 - Emissions
 - Atmospheric chemistry
- Deposition budget 2002-2004 versus 2010-2012
- Future outlook and conclusions



Reduced N in the Environment

- NH₃ is the primary atmospheric base
 - Precursor to atmospheric particulate matter formation
 - Deleterious to human respiratory and cardiovascular systems
 - Short term climate forcer
- Reduced nitrogen deposition accounts for ~35% of the total nitrogen deposition in the U.S.
 - Contributes to excess nitrogen in ecosystems
 - Surface water eutrophication and terrestrial biodiversity loss
 - Contributes to soil and surface water acidification
- NH₃ air-surface exchange is bi-directional
 - Can be emitted (evasion) or deposited
 - Depends on land use, environmental variables, ambient NH₃
 concentration and land management practices

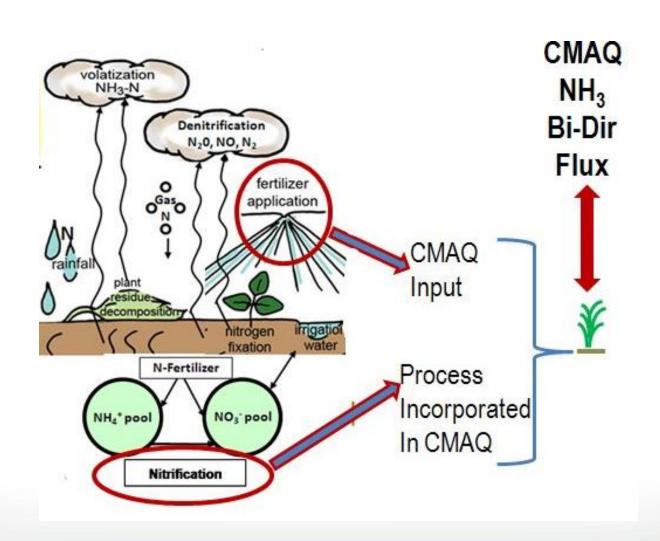


Advancements in CMAQ

- Bidirectional exchange
 - Coupled Agro-ecosystem model to the chemical transport model
 - Environmental Policy Integrated Climate (EPIC) model
 - Couples agricultural cropping management and soil geochemical processes with CMAQ
 - Dynamic NH₃ emissions from fertilizer application
 - Dependent on fertilizer composition, weather, soil conditions, crop, application method, etc.
- Temporal CAFO NH₃ emissions
 - Applies physical constraints for hourly emissions estimates from annual totals submitted by the states
- Improvements in modeled convective precipitation
 - Reduces the overestimate of precipitation in the summer months.

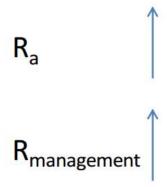


Bidirectional NH₃ Exchange





Improvements in CAFO Emissions



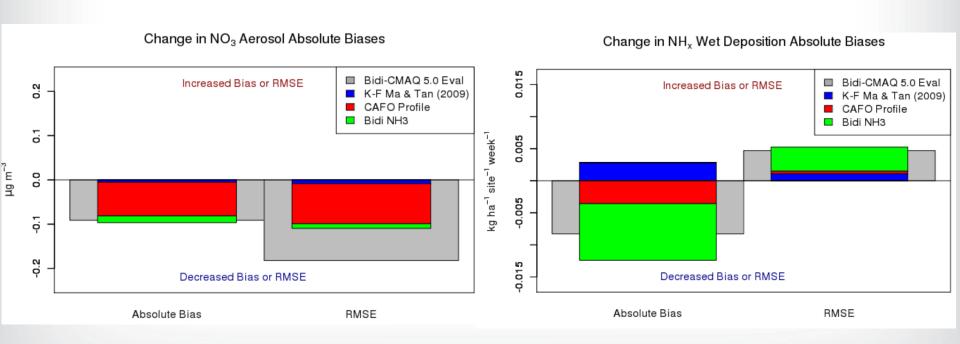
- Use a resistance model approach
- Assume NH₃ at the source is >> that in atmosphere
- Assume NH₃ emissions originate from an aqueous pool
- Key parameters are atmospheric resistance and NH₄ (aq) ← NH_{3(g)} equilibrium





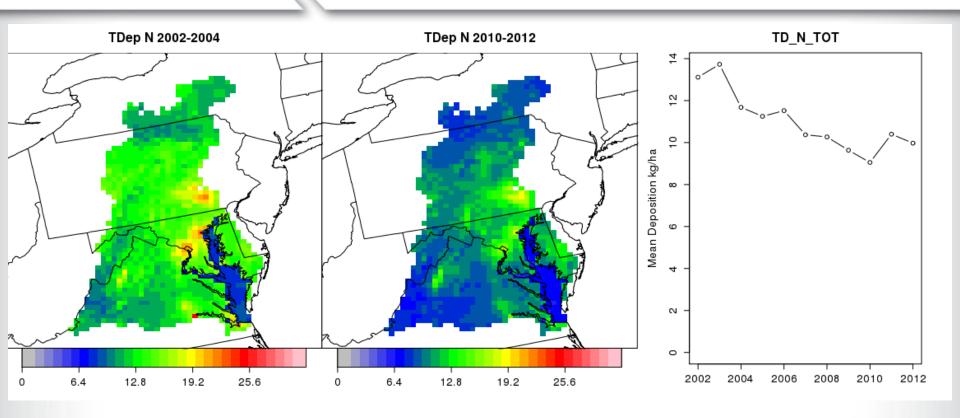
CMAQ Improvements

- Bidirectional exchange and CAFO updates significantly improve model evaluations against NHx wet deposition and NO₃ ambient aerosol observations
- Further evaluation in the next presentation by Robin Dennis





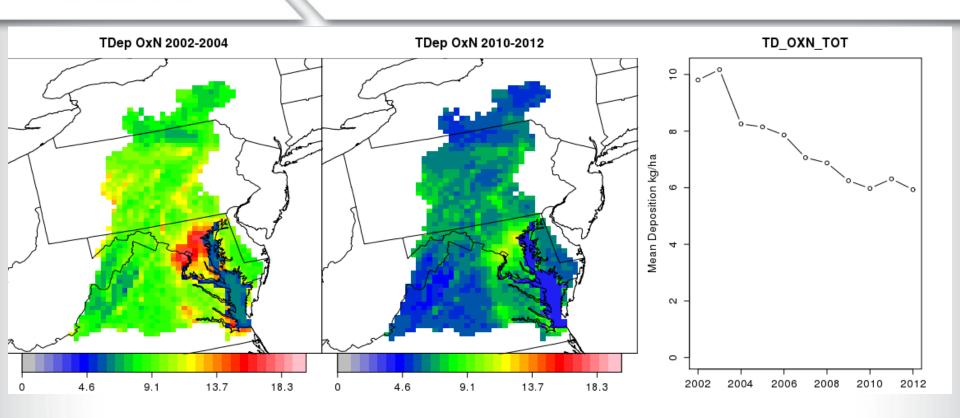
Nitrogen Total Deposition



- 24% reduction in total nitrogen atmospheric deposition
- Clear benefits from air-quality regulations



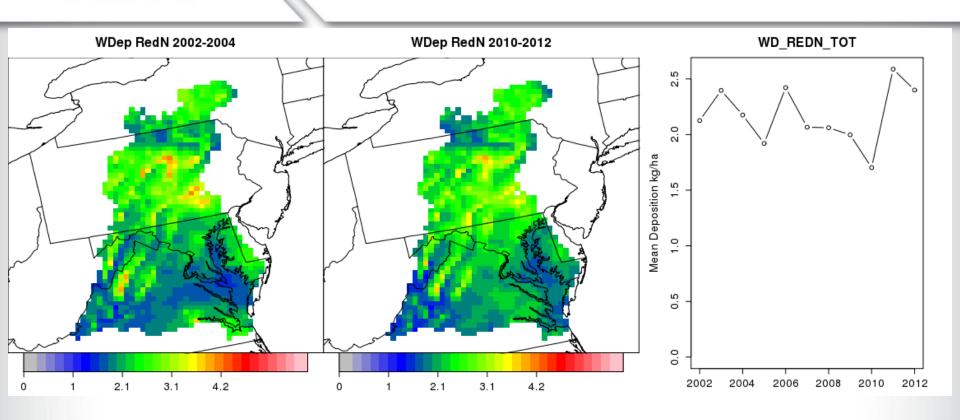
Oxidized Nitrogen Total Deposition



- 35% reduction in oxidized nitrogen atmospheric deposition
- Closely follows NOx emission reductions



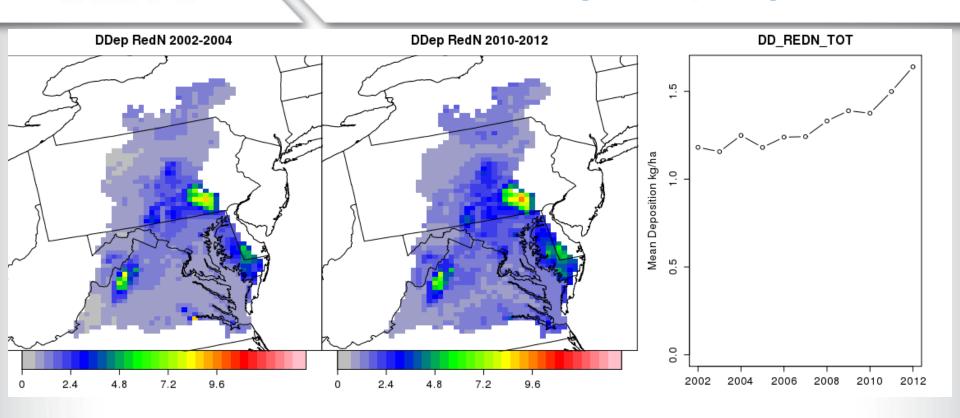
Reduced Nitrogen Wet Deposition



- No trend in reduced nitrogen wet deposition
- No to slightly increasing trend in NH₃ emissions
- Variability driven by precipitation



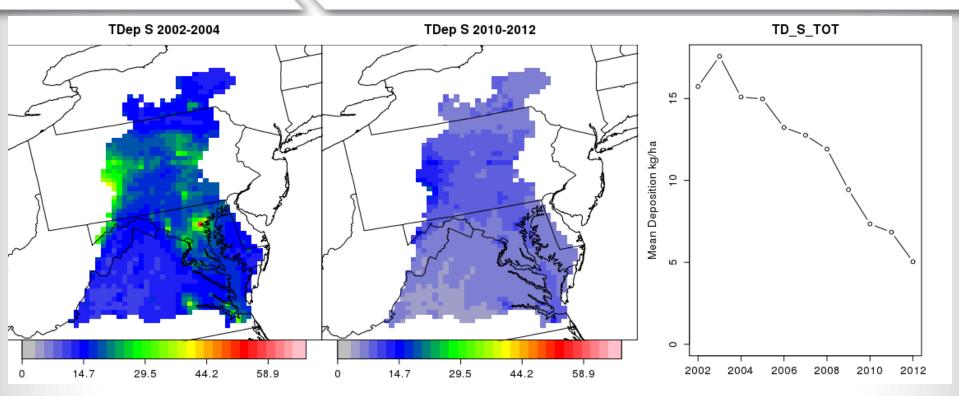
Reduced Nitrogen Dry Deposition



- 26% increase in reduced nitrogen dry deposition
- No to slightly increasing trend in NH₃ emissions
- Trend driven by NOx and SOx emission reductions (aerosol partitioning)



Sulfur Total Deposition



- 60% reduction in sulfur deposition
- Reflects large reductions in SO₂ emissions

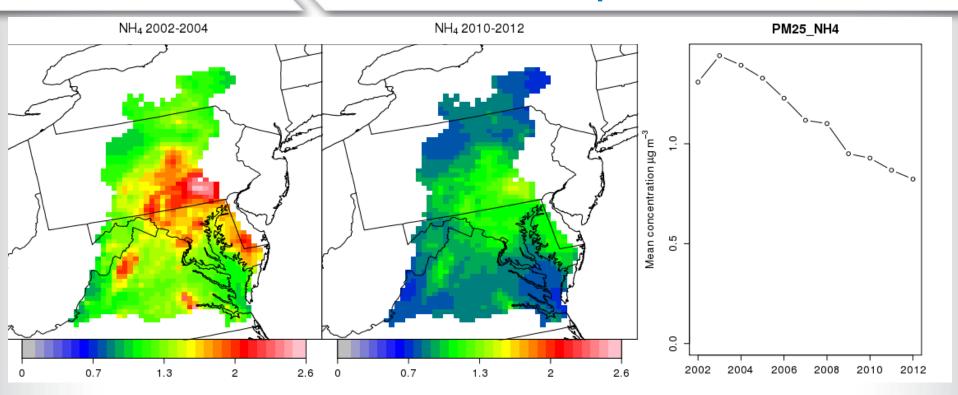


Atmospheric NH₃

- NH₃ neutralizes SO₄ and NO₃ aerosols to form NH₄
 aerosols as ammonium sulfate and ammonium nitrate
- Gaseous NH₃ and aerosol NH₄ behave differently in the atmosphere
 - NH₃ dry deposits more rapidly than NH₄
- NO_x and SO_x emission reductions
 - Lowered SO₄ and NO₃ aerosol concentrations
 - Results in lower NH₄ aerosol concentrations
- Static NH₃ emissions with these reductions results in more ambient gaseous NH₃
 - More local NH₃ deposition
 - Reduction of long range reduced nitrogen transport



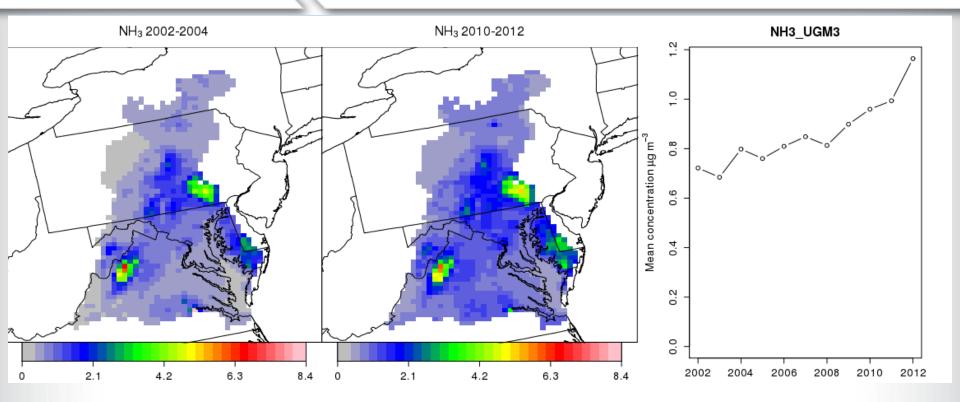
Ambient NH₄ Aerosol



- 37% reduction in ambient NH₄ aerosol concentration
- Reflects reductions in SO₂ and NO_x emissions



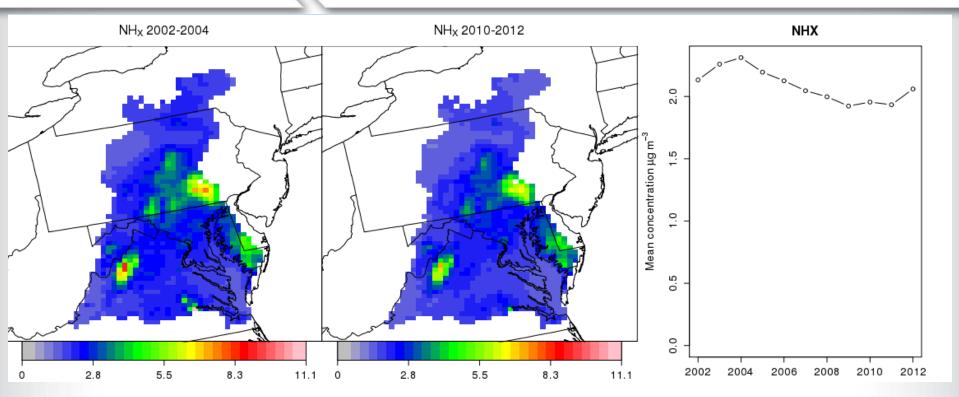
Ambient Gaseous NH₃



- 41% increase in ambient NH₃ gaseous concentration
- Due to reduction in aerosol partitioning
- Almost the inverse of the NH₄ reduction



Ambient NH_x (NH₃+NH₄)



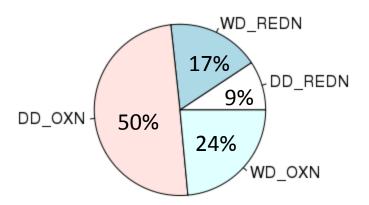
- 11% reduction in ambient NH_x
- Static NH₃ emissions
- Reduction is due to increase in NH₃ dry deposition



N Deposition Budget

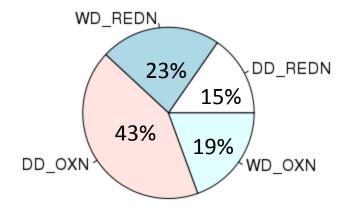
2002-2004

Mean: 13 kg N/ha



2010-2012

Mean: 10 kg N/ha



- Overall N deposition has decreased due to regulations
- Ratio of oxidized to reduced N deposition is changing
- Oxidized N deposition is decreasing
 - Due to controls on combustion sources
- Reduced N deposition is increasing
 - Due to changes in atmospheric composition and a lack of controls on NH₃ emissions



Conclusions and Future Outlook

- CMAQ captures the NOx regulations reductions in N deposition
- Oxidized nitrogen deposition primarily from combustion sources has decreased more rapidly than total N deposition
- Reduced nitrogen deposition primarily from agricultural sources has increased due to atmospheric composition changes
 - Gaseous NH₃ is deposited more rapidly than aerosol NH₄
- Future reductions in NOx will likely plateau
- Reduced N deposition will likely represent a larger portion of the deposition budget
- Mitigating NH_x loading will be tricky
 - Need to consider run off and ground water infiltration
 - A linked agriculture, air-quality, hydrology and water quality modeling system is needed