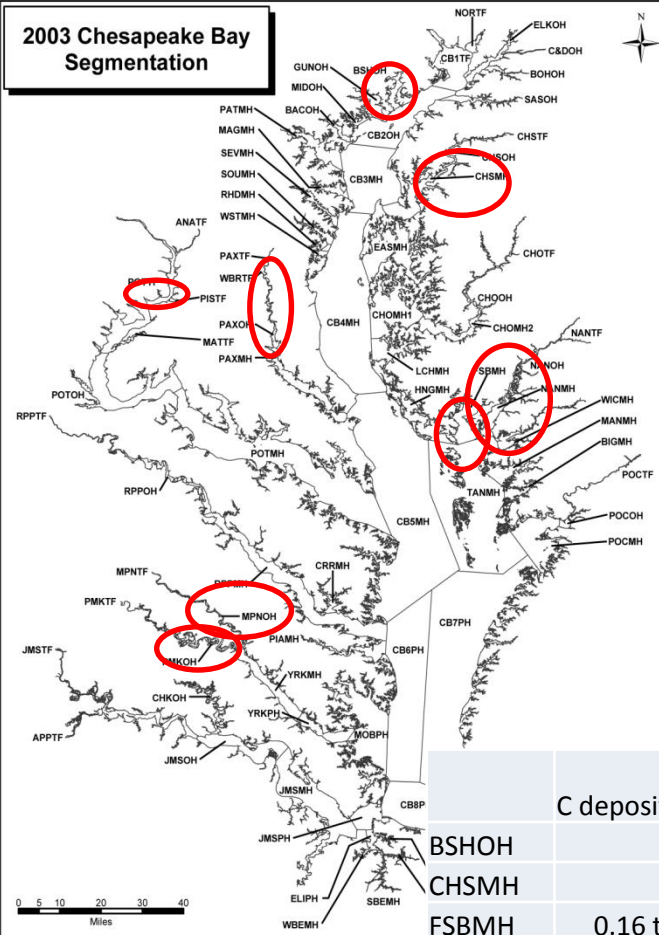


2003 Chesapeake Bay Segmentation



Hot Spots for Calibration

	C deposition	N deposition	P deposition	denitrification	solids deposition	respiration
BSHOH		0.008 to 0.032	0.001 to 0.006			
CHSMH		0.02 to 0.064	0.01 to 0.019		3.6	
FSBMH	0.16 to 0.33				0.3	
MPNOH	0.24 to 2.77	0.019 to 0.238	0.004 to 0.085		1.43 to 42.0	
MPNTF						
NANMH	0.033 to 0.126				1.61 to 8.12	
NANOH	0.033 to 0.126				1.61 to 8.12	
PAXOH		0.008	0.002		5.75	
PAXTF		0.033 to 0.064	0.01	0.108 to 0.197	5.75	
PMKOH	0.61	0.05		0.04		1.12 to 2.77
POTTF	1.44			0.043 to 0.06	5.88	
WICMH	0.033 to 0.126	0.037	2.74x10 ⁻⁵ to 0.004		1.61 to 8.12	
CHOMH		0.053 to 0.074	4.9x10 ⁻⁴ to 0.005			
WQGIT			0.0016	0.026		

Wetlands Module

- We don't want to develop a complete wetlands biogeochemical model.
- We do want to develop a simplified module that includes:
 - Particle burial (organic and inorganic)
 - Respiration
 - Denitrification
 - Primary production?
 - Others?

Particle Settling

$$V \cdot \frac{dC}{dt} = \textit{Transport} + \textit{Kinetics} - WSw \cdot C \cdot Aw$$

V = volume of WQM cell adjacent to wetlands

C = concentration

WSw = wetland settling velocity

Aw = area of wetland adjacent to WQM cell

This applies to all particles, organic and inorganic. Present settling rates 0.05 m/d for most particles, 0.005 m/d for phytoplankton.

Respiration

$$V \cdot \frac{dC}{dt} = \text{Transport} + \text{Kinetics} - f(DO) \cdot f(T) \cdot WOC \cdot A_w$$

V = volume of WQM cell adjacent to wetlands

C = concentration

f(DO) = limiting factor = $DO / (K_h + DO)$

f(T) = temperature effect

WOC = wetland oxygen consumption

A_w = area of wetland adjacent to WQM cell

At present, WOC = 0.5 g DO/sq m/d at 20C. WOC doubles for a 10C temperature increase. K_h = 1.0 g DO/m³.

Previous calibration had WOC = 1 g DO/sq m/d and no limiting factor. Wetland areas from TMDL model.

Denitrification

$$V \cdot \frac{dC}{dt} = \text{Transport} + \text{Kinetics} - \text{MTC} \cdot f(T) \cdot C \cdot A_w$$

V = volume of WQM cell adjacent to wetlands

C = nitrate concentration

MTC = mass-transfer coefficient

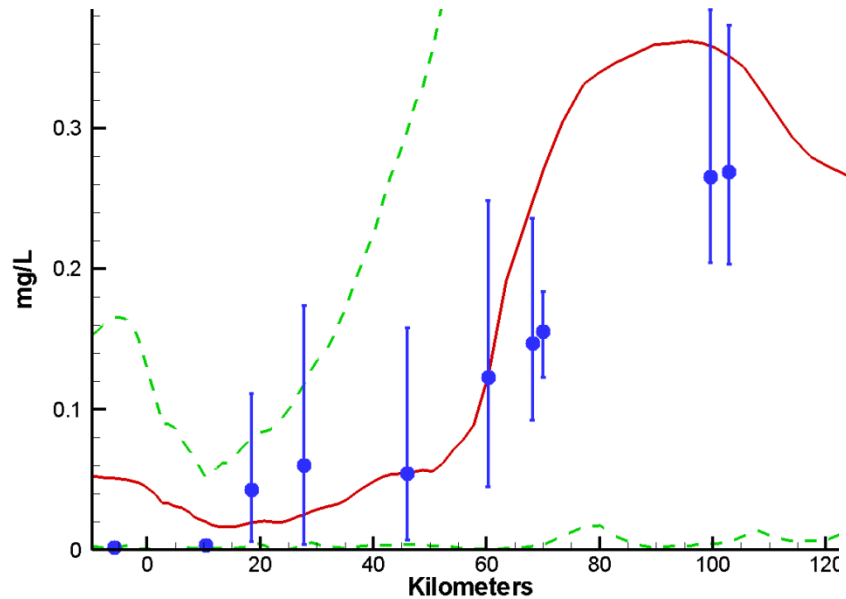
f(T) = temperature effect

A_w = area of wetland adjacent to WQM cell

At present, the mass-transfer coefficient is 0.05 m/d.

Denitrification doubles for a 10C temperature increase.

York River 2002-2011 Run71
Surface Nitrate Summer 2004

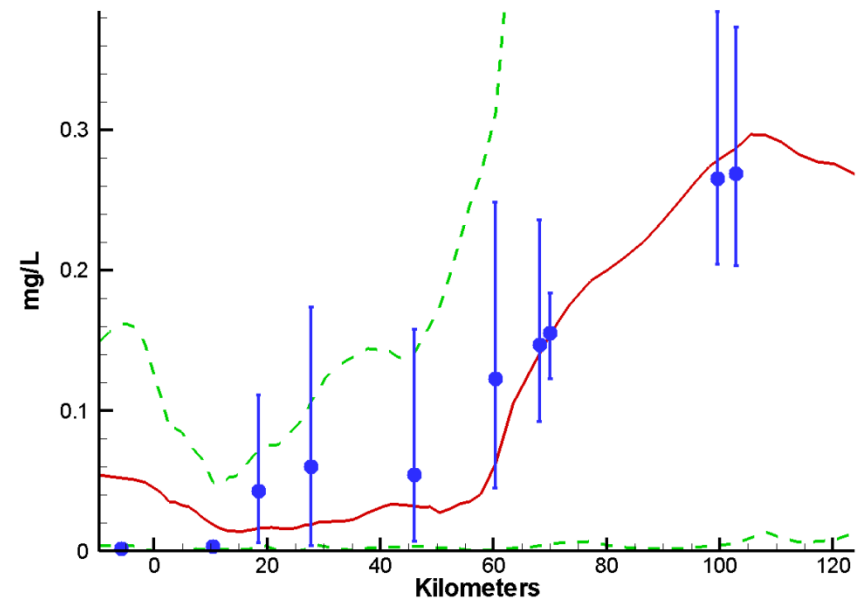


Nitrate in York River

No Wetlands

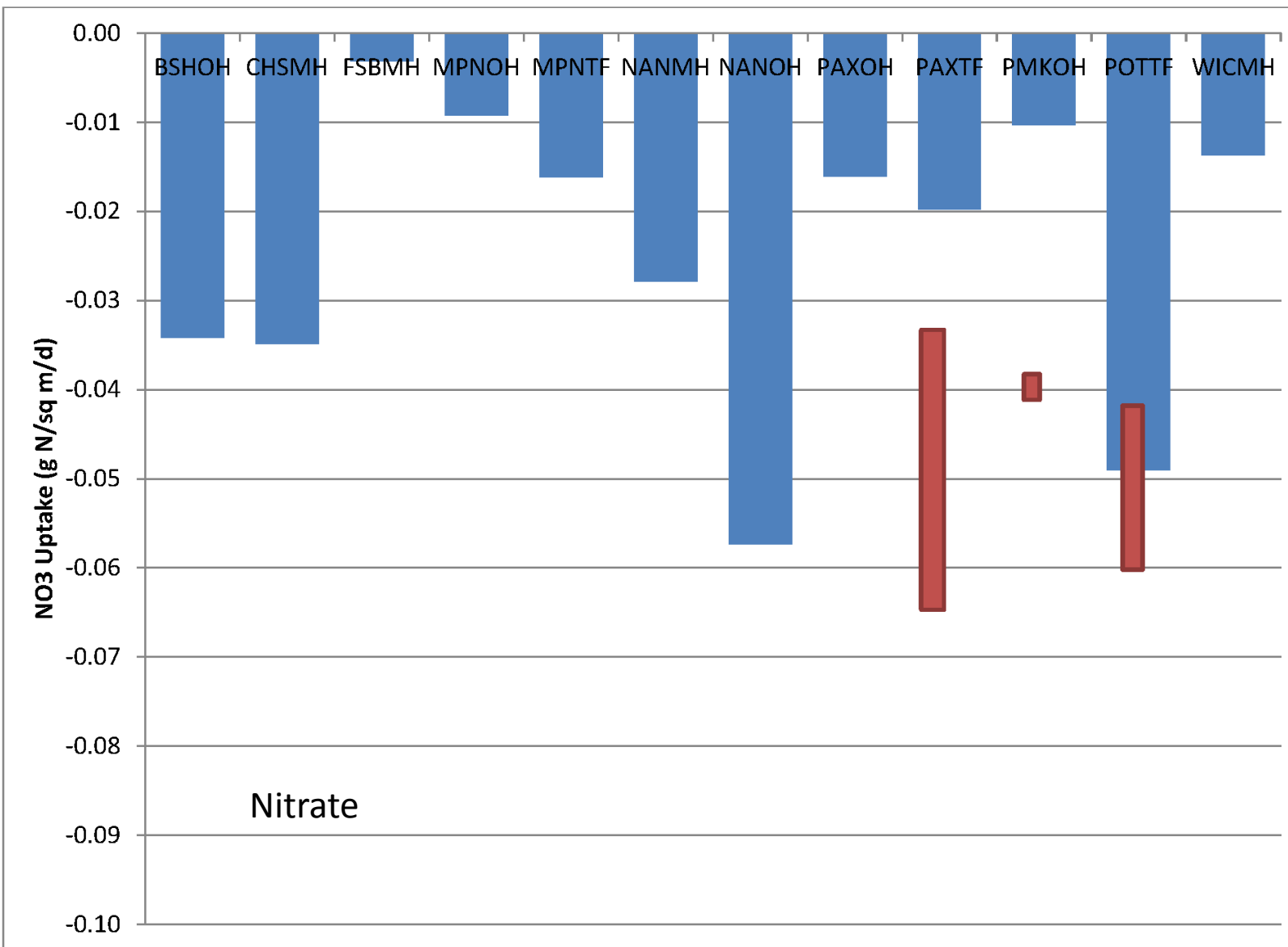


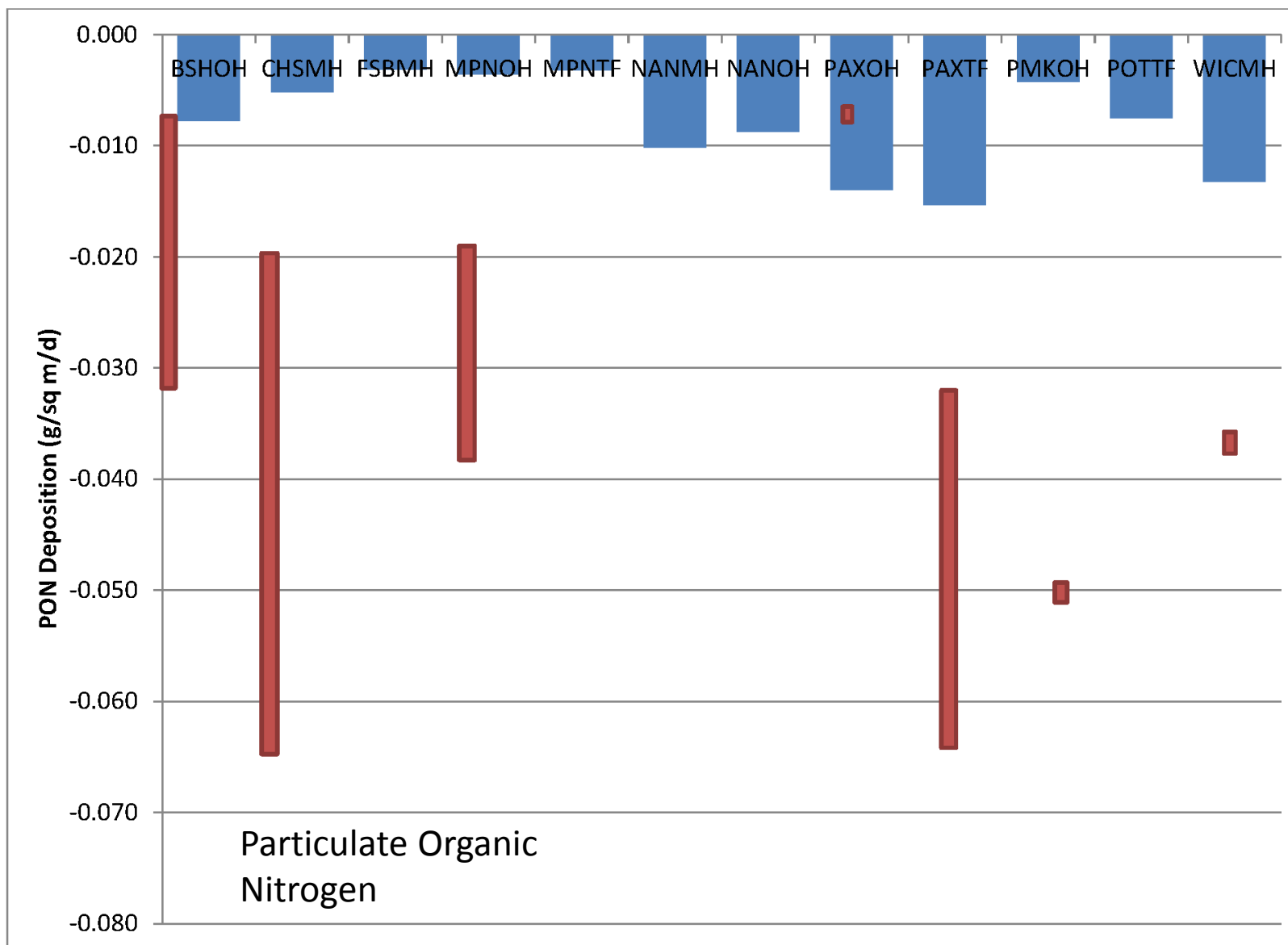
York River 2002-2011 Run84
Surface Nitrate Summer 2004

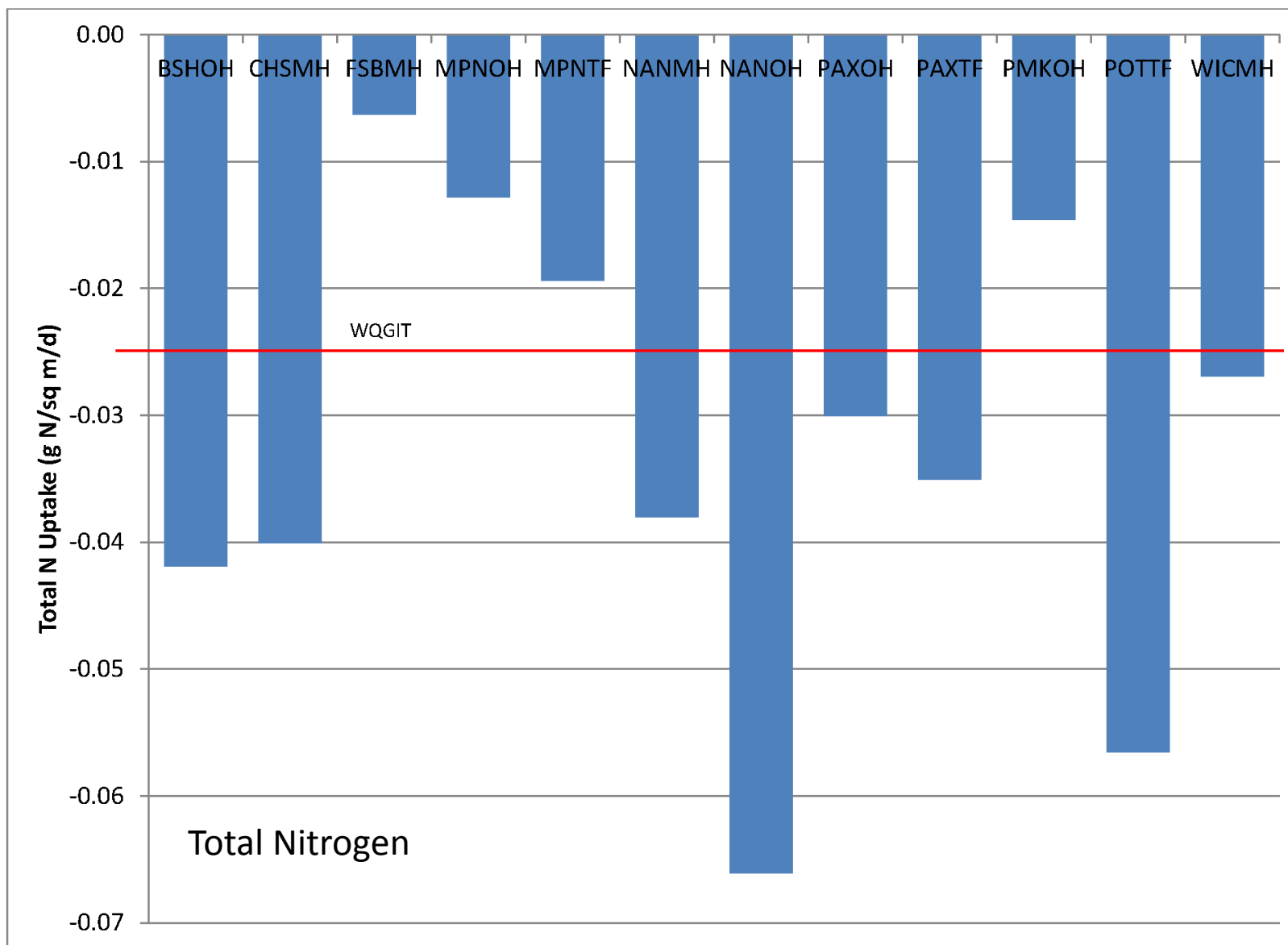


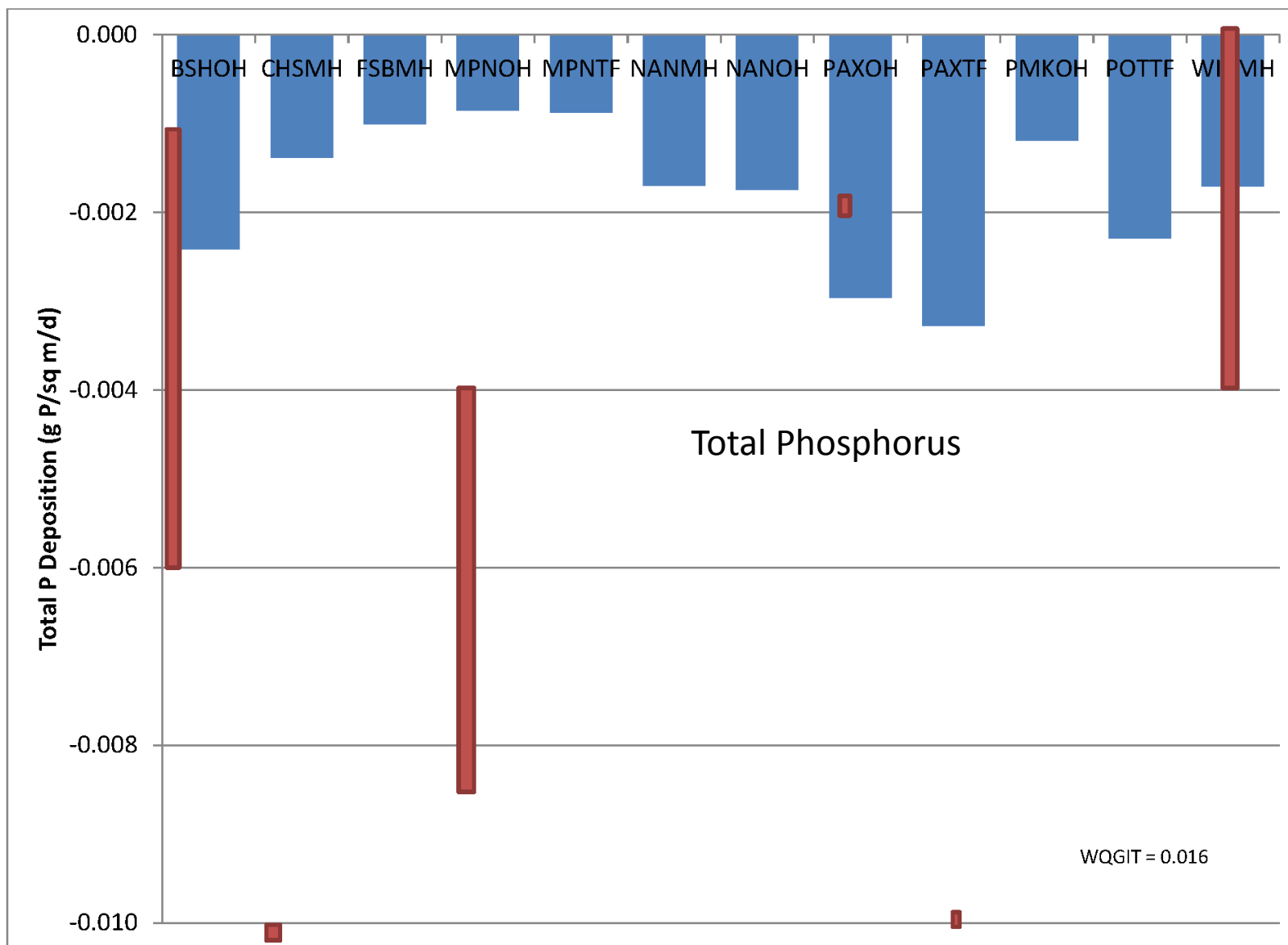
With Wetlands

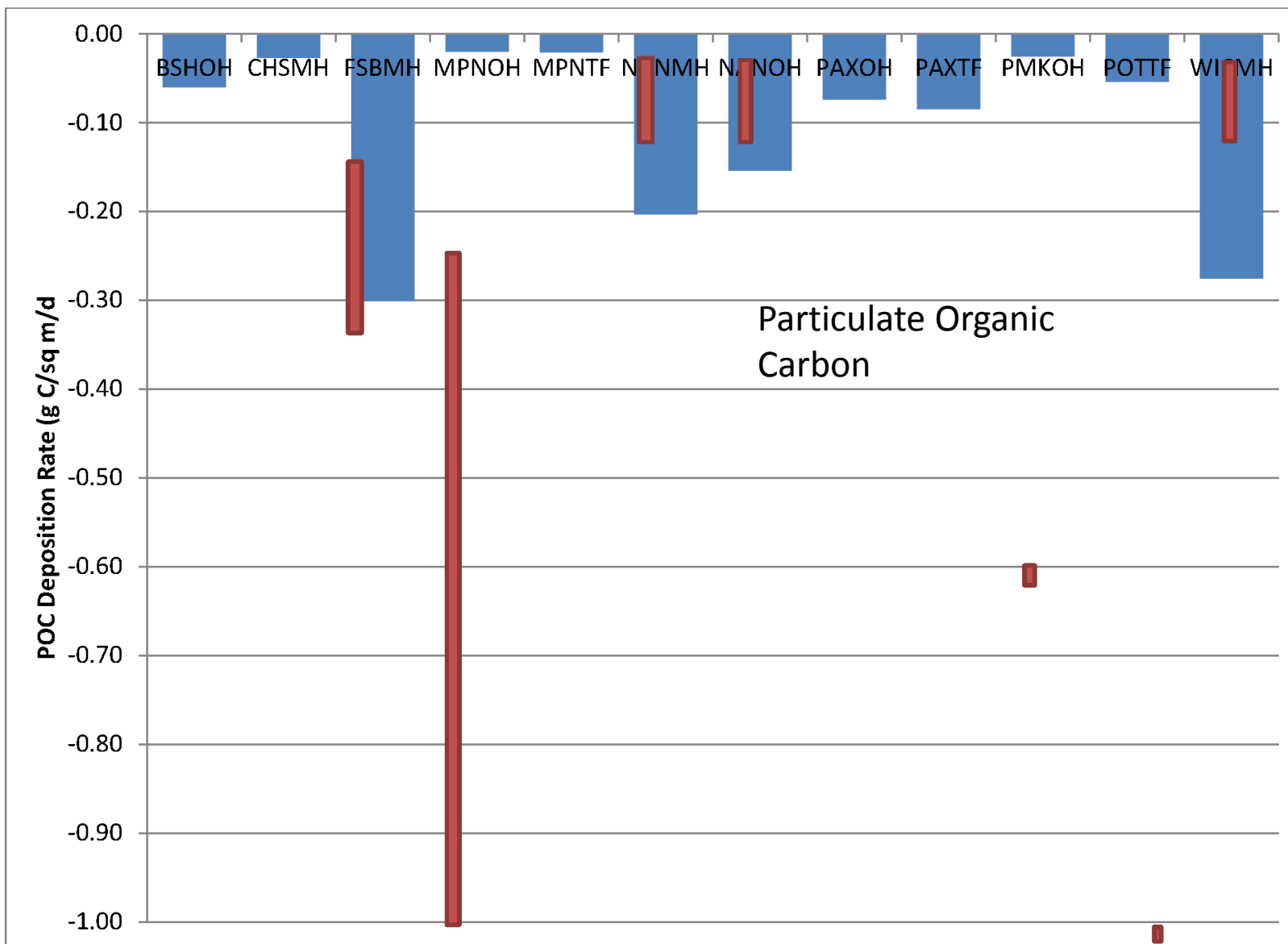


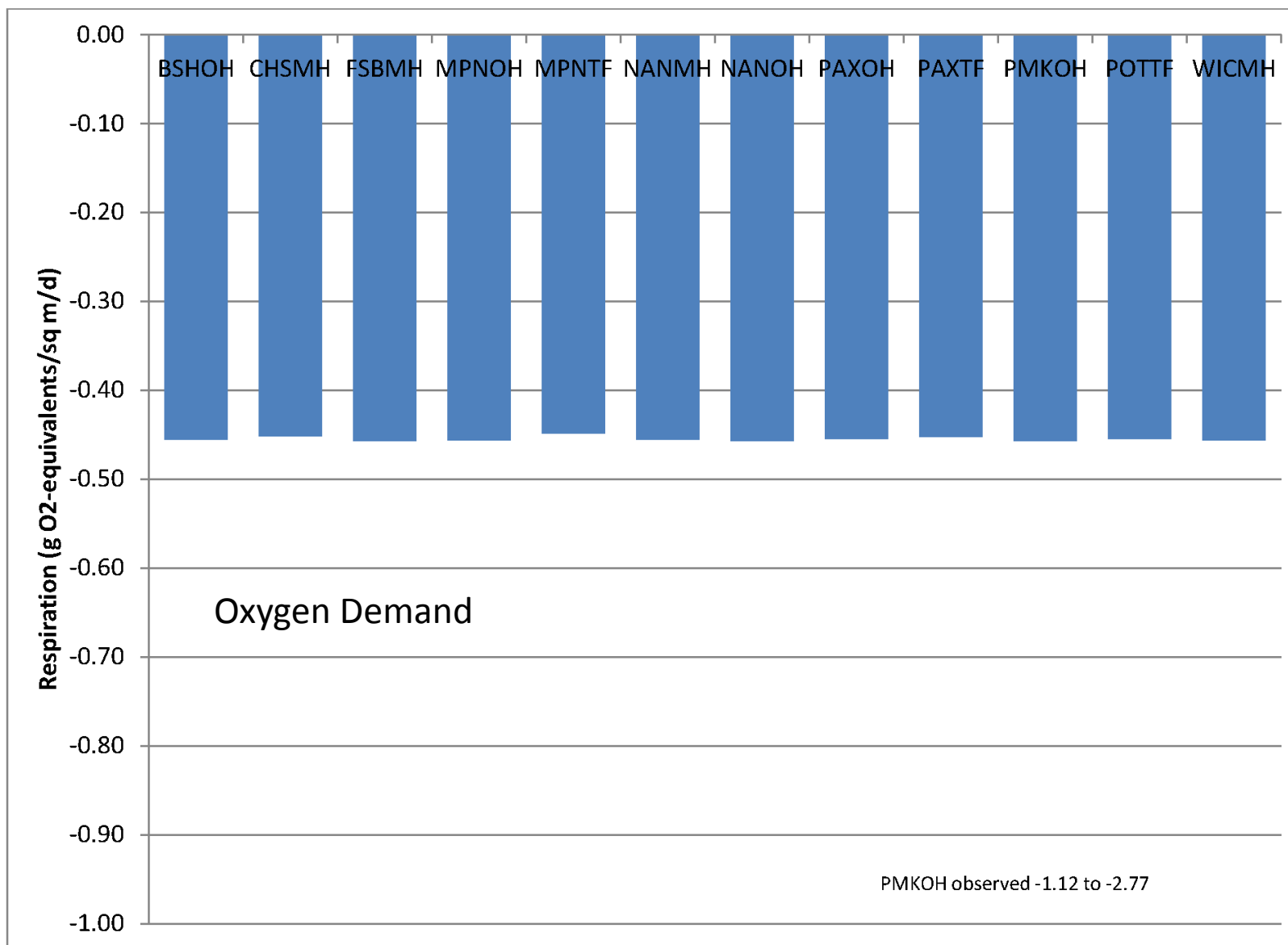


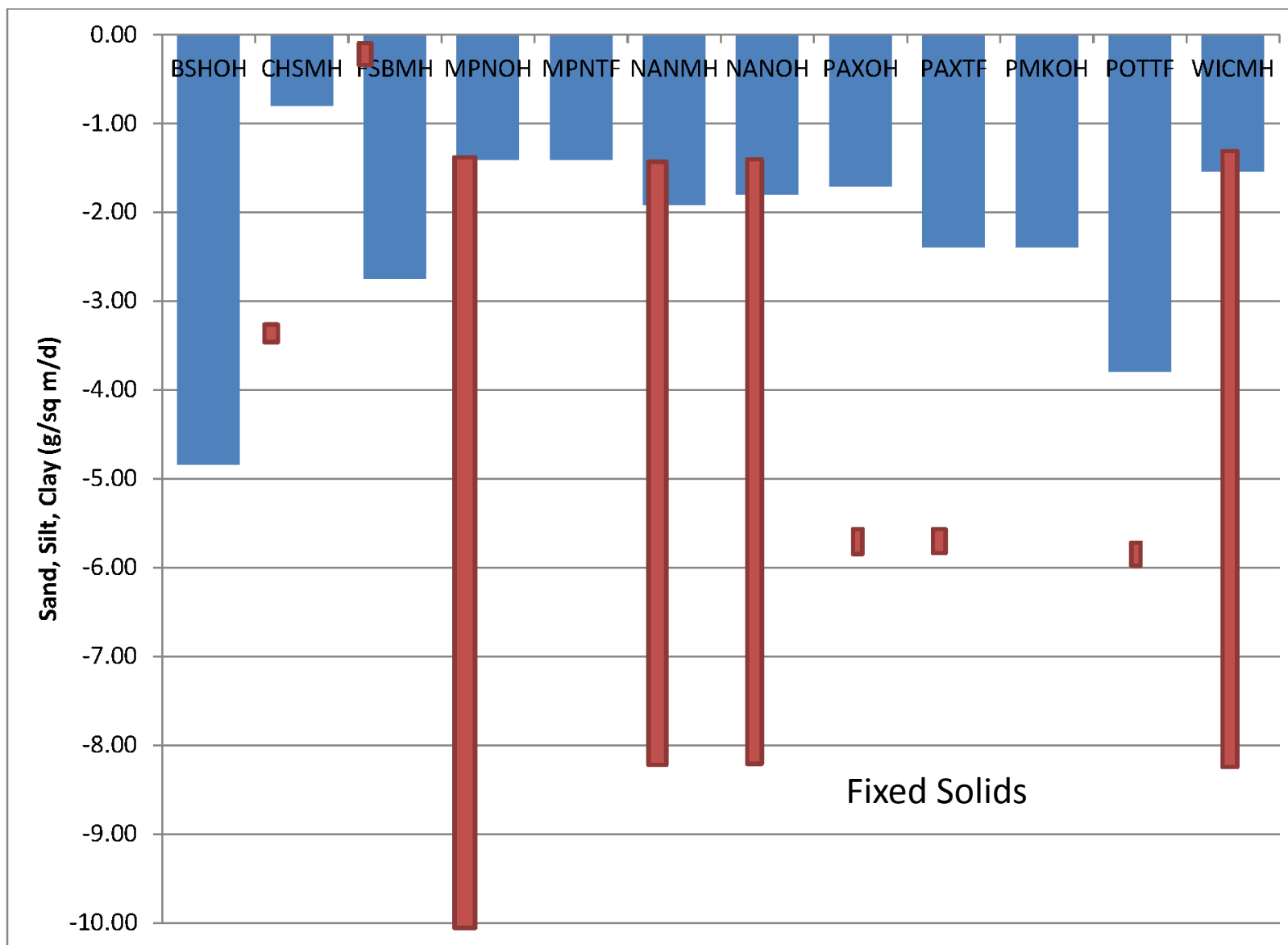


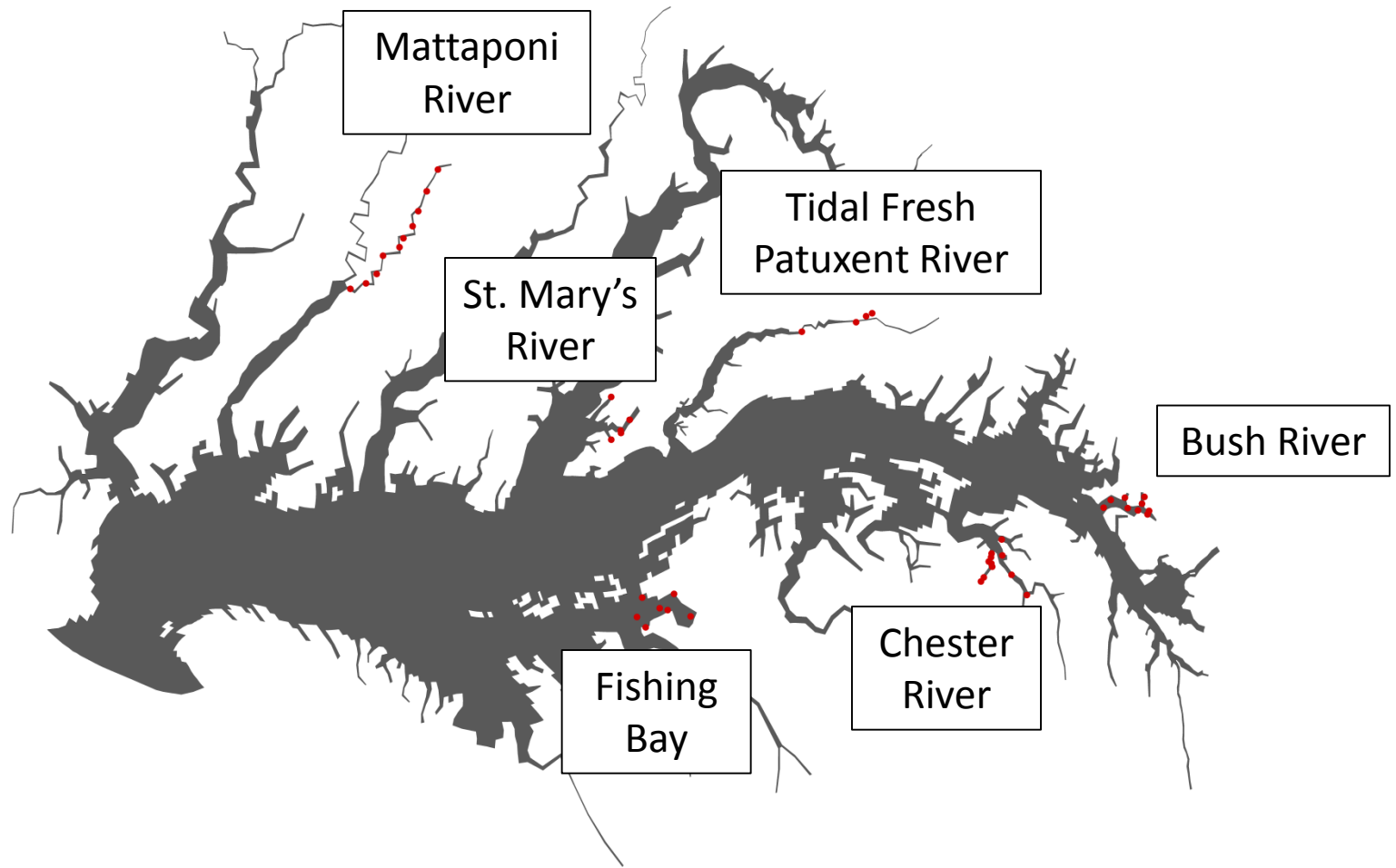




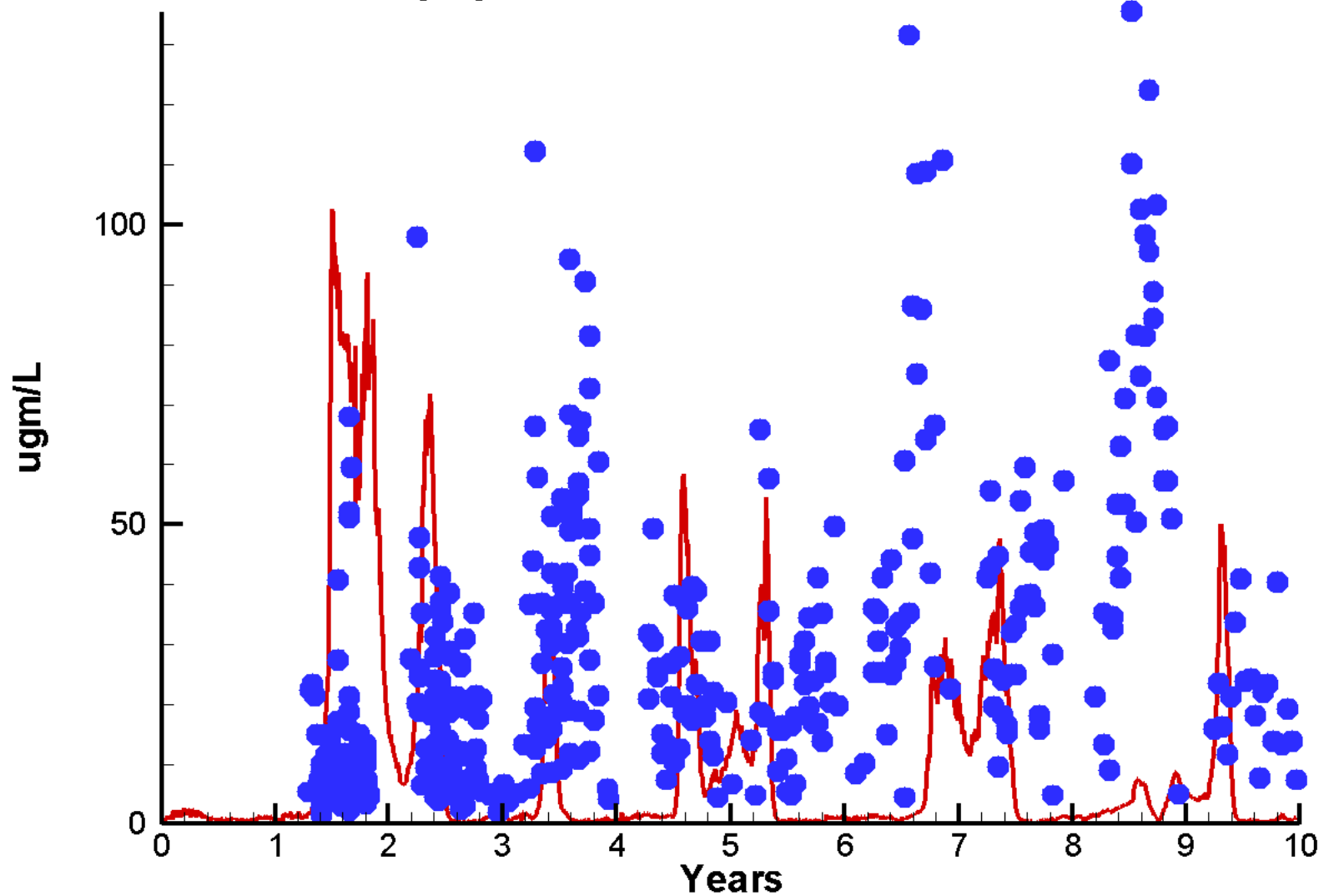






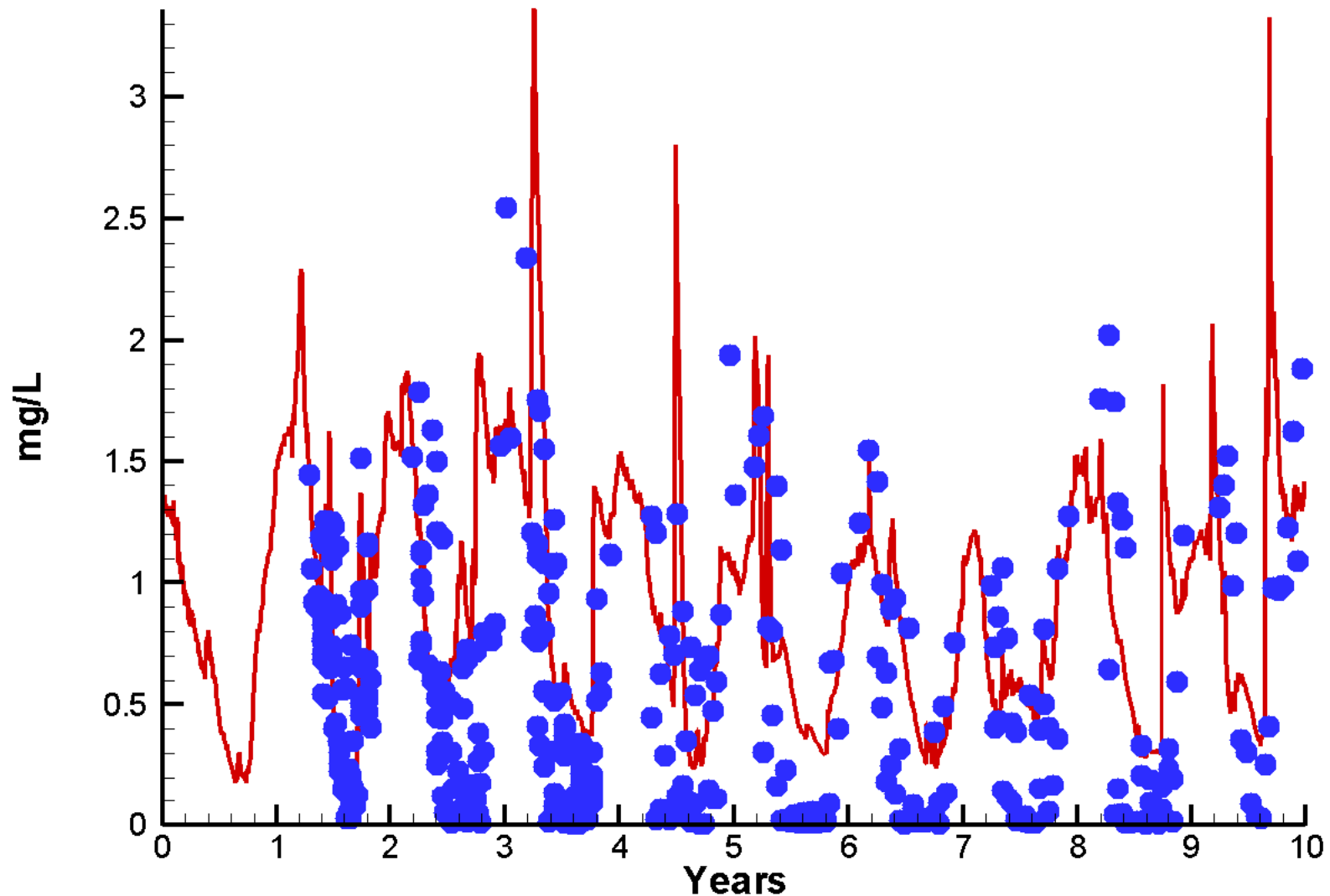


Run119 2002-2011 Chlorophyll Bush River



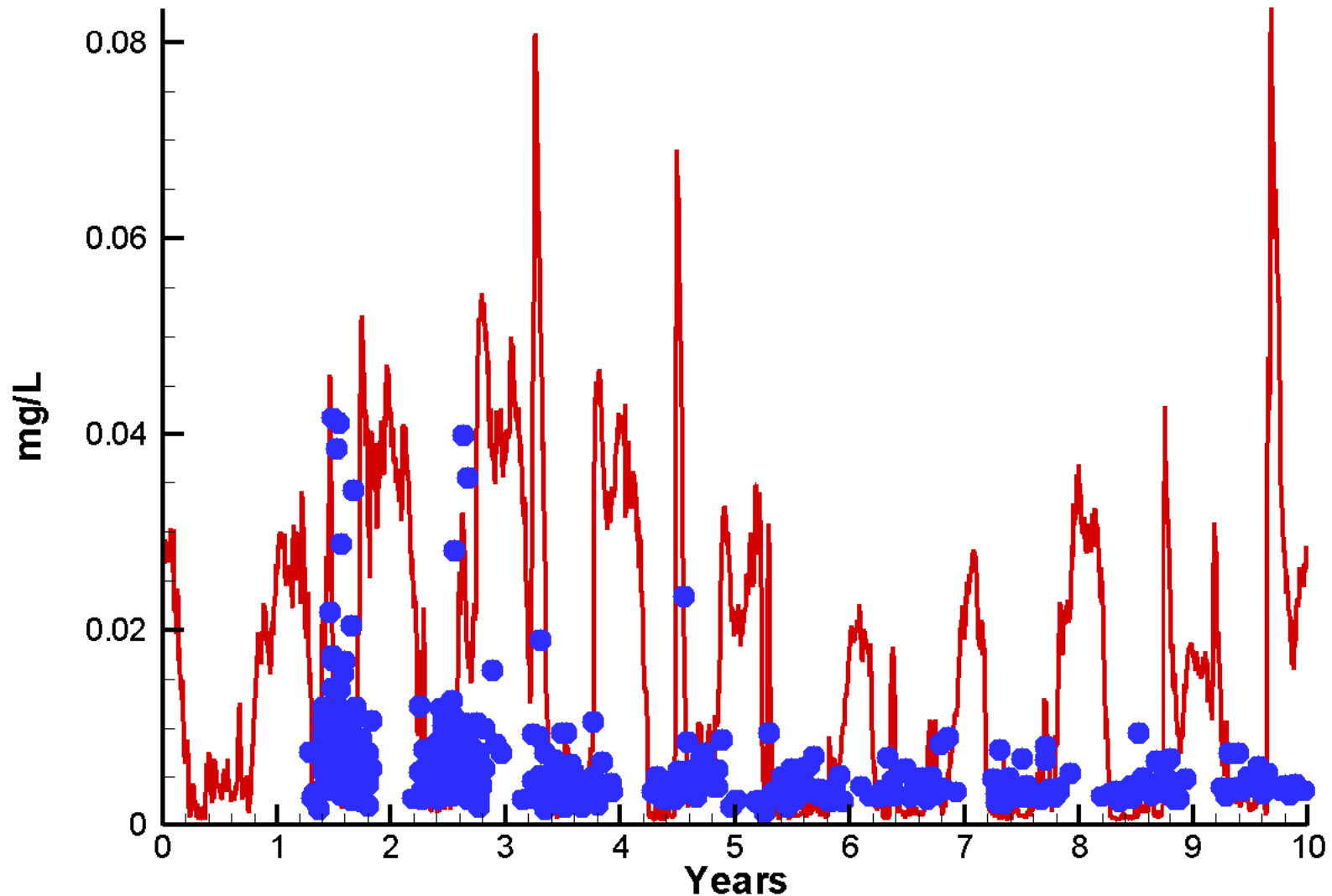
Run119 2002-2011

Dissolved Inorganic Nitrogen Bush River

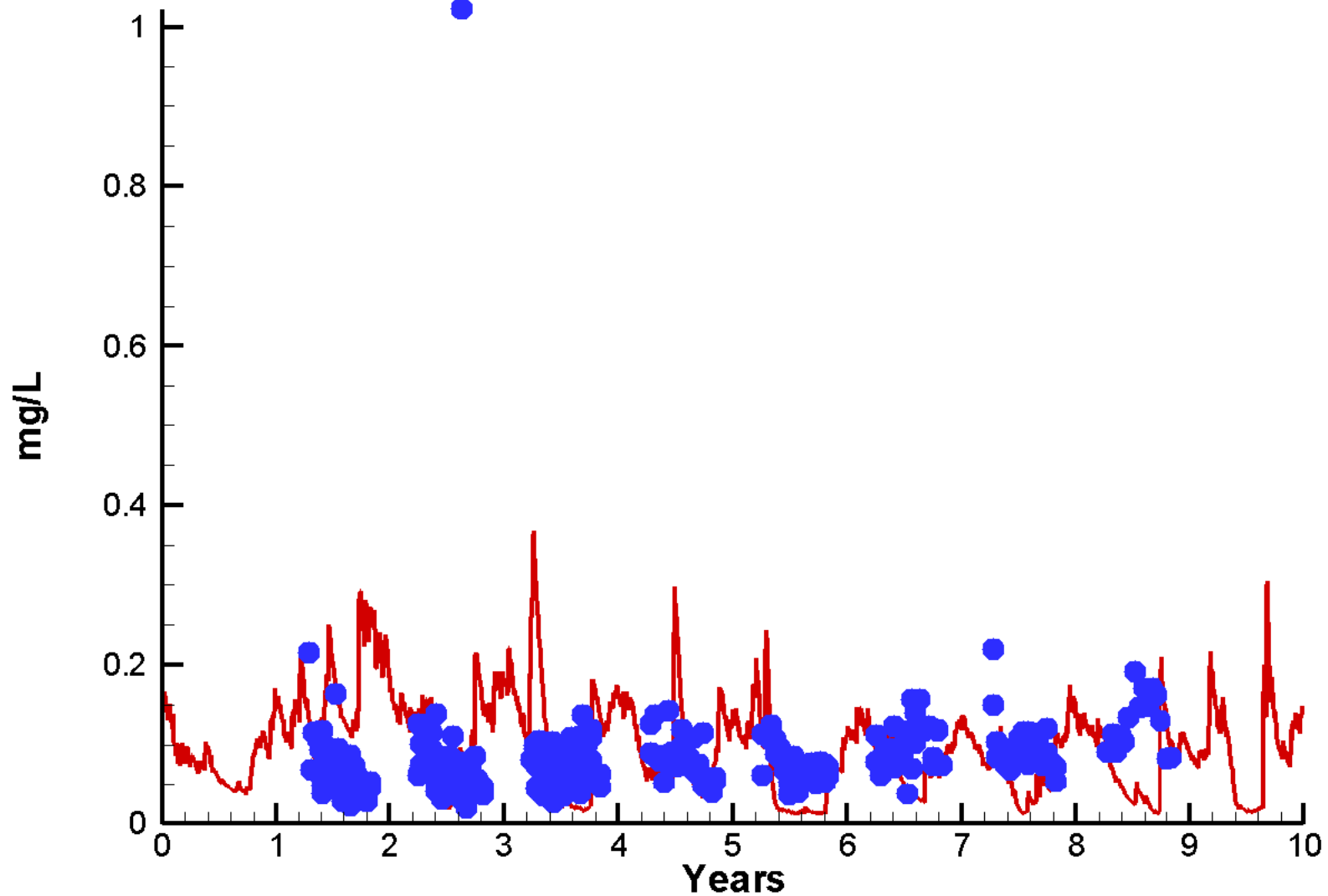


Run119 2002-2011

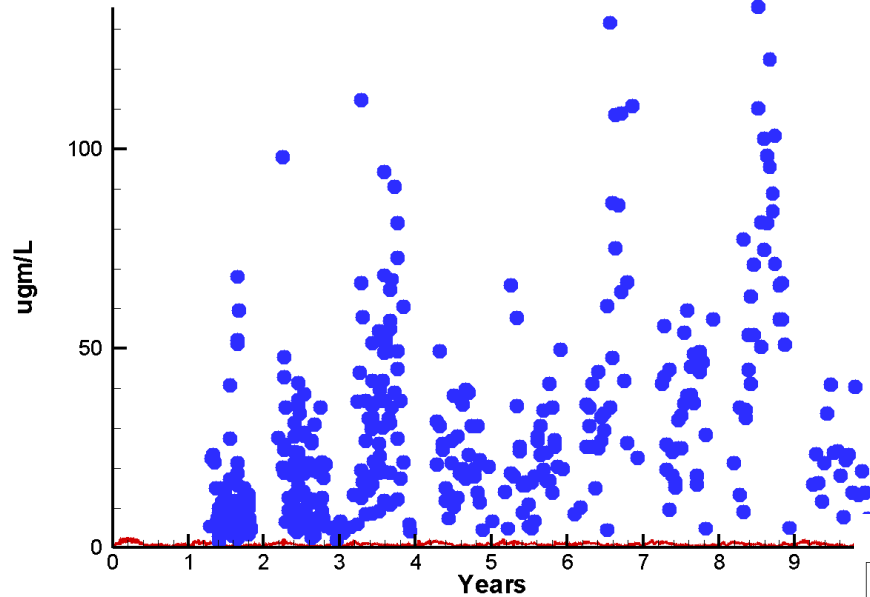
Dissolved Inorganic Phosphorus Bush River



Run119 2002-2011
Total Phosphorus Bush River



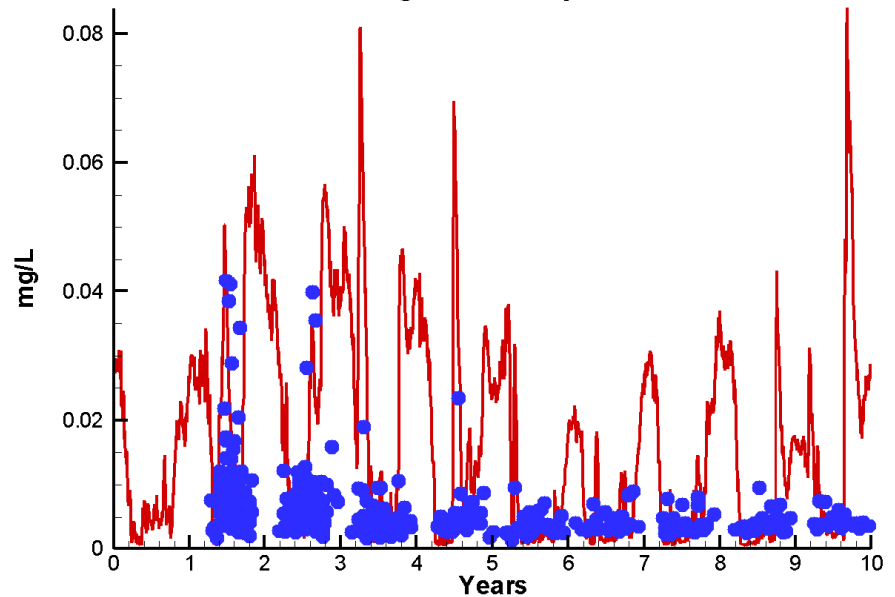
Run117 2002-2011
Chlorophyll Bush River



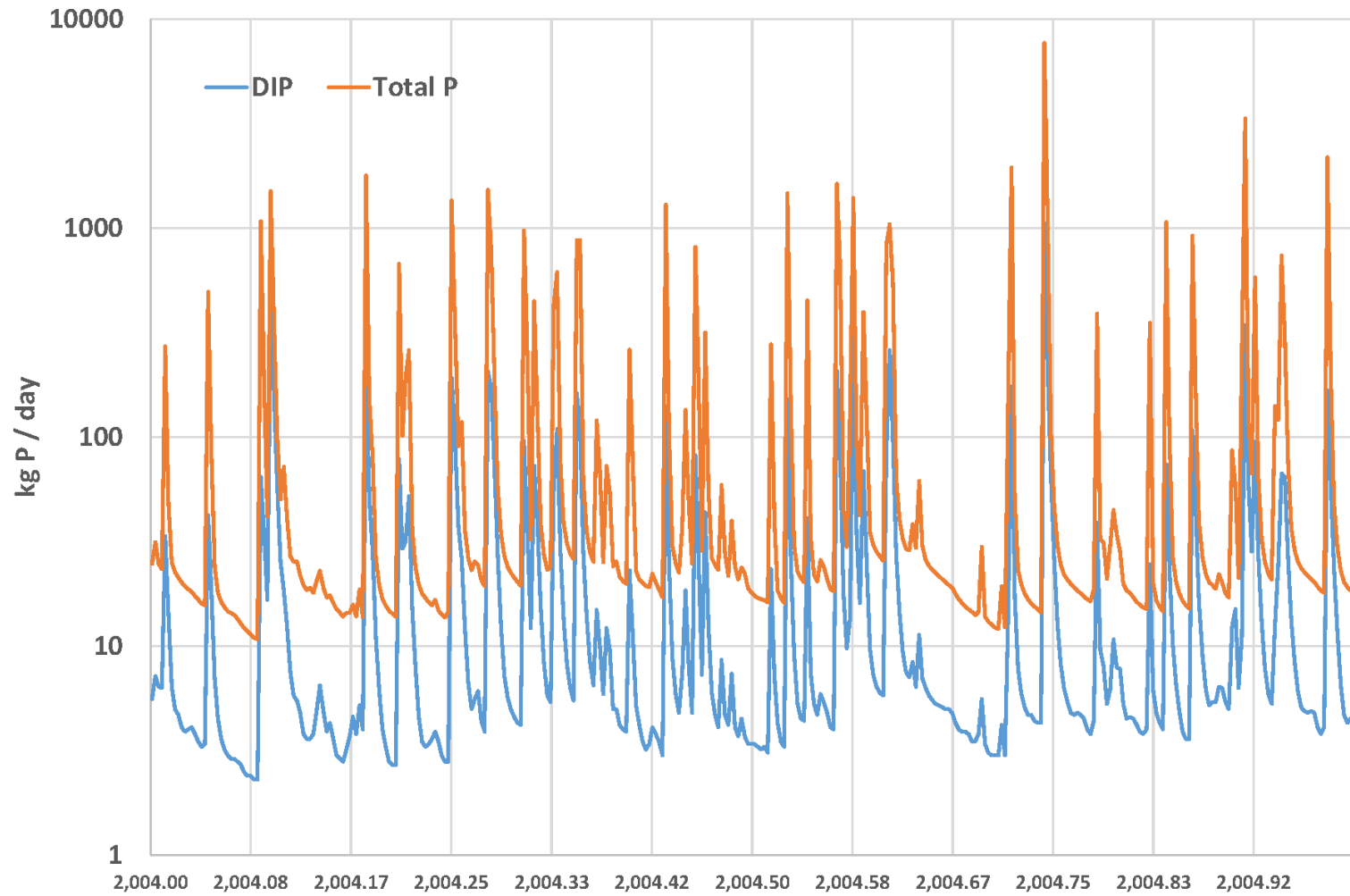
This was a bad model run. We grew no phytoplankton.

Even with no algal uptake, we have no DIP.

Run117 2002-2011
Dissolved Inorganic Phosphorus Bush River

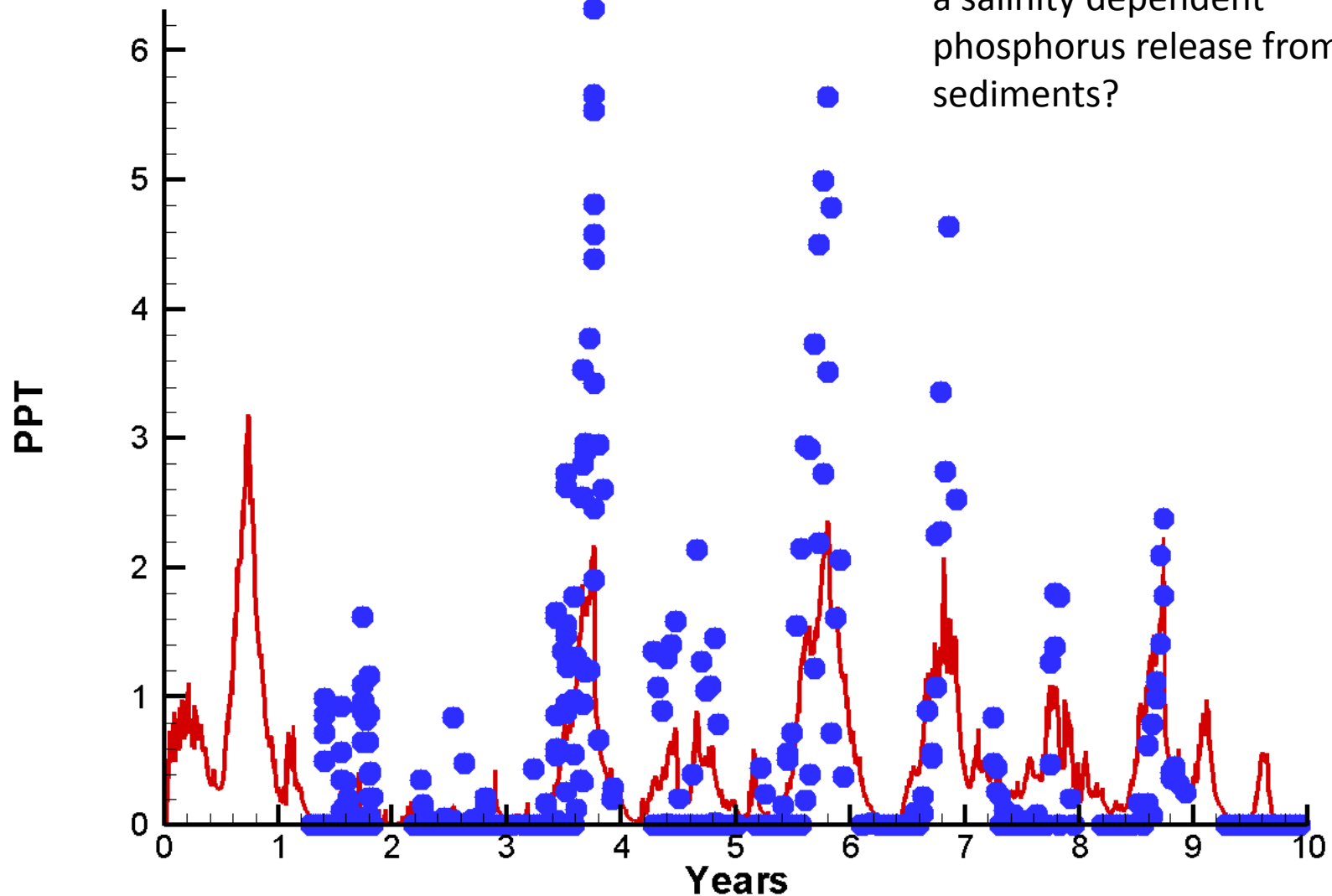


2004 Bush River P Loads

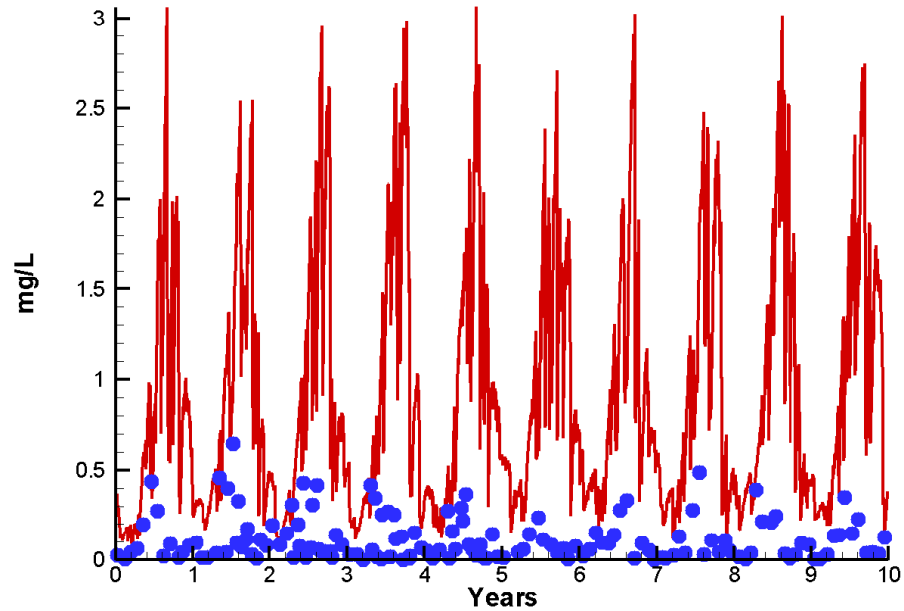


Run119 2002-2011 Salinity Bush River

Possible Saving Grace – Is there
a salinity dependent
phosphorus release from
sediments?

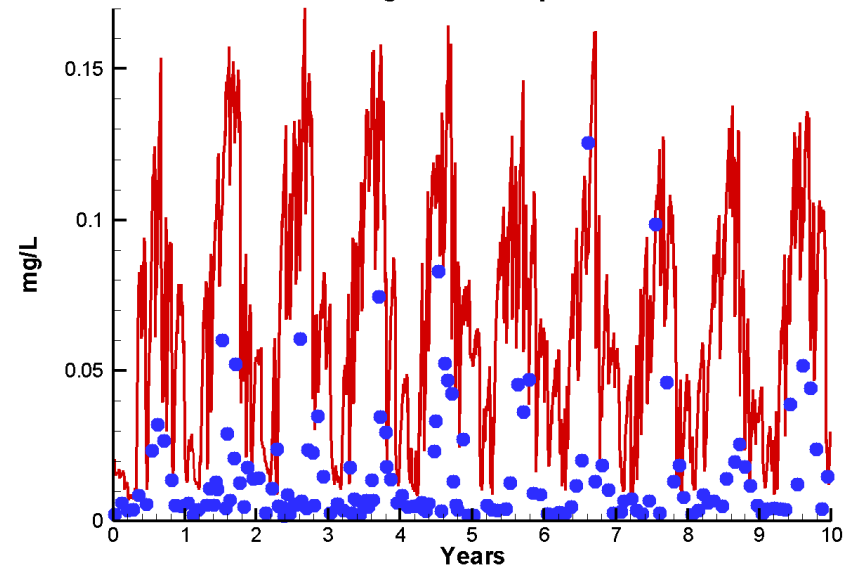


Run117 2002-2011
Ammonium ET4.2 Bottom

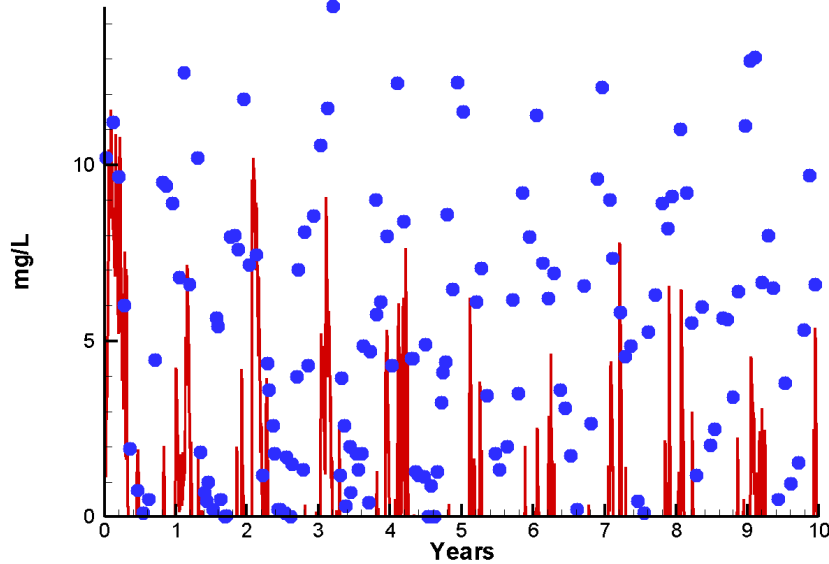


Let's move to the
Chester River, Station
ET4.2

Run117 2002-2011
Dissolved Inorganic Phosphorus ET4.2 Bottom



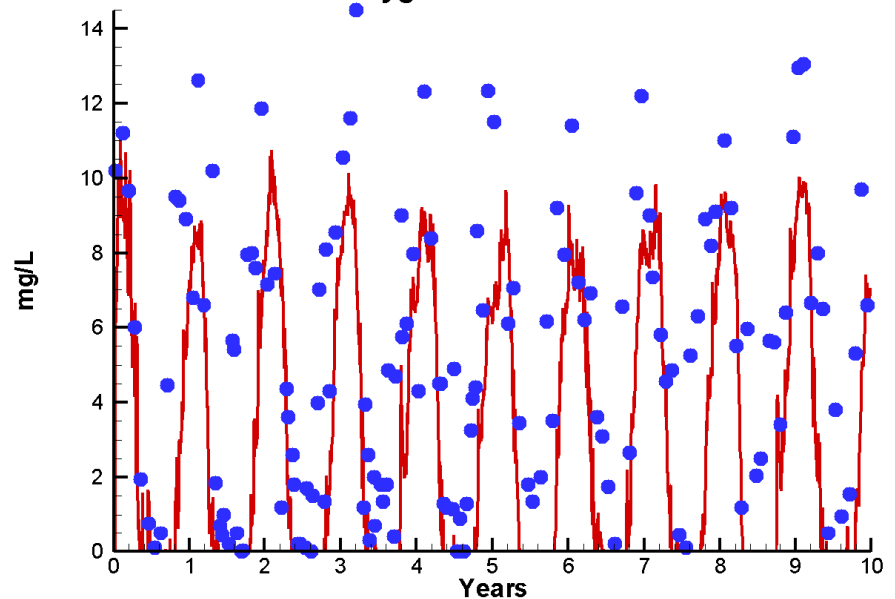
Run117 2002-2011
Dissolved Oxygen ET4.2 Bottom



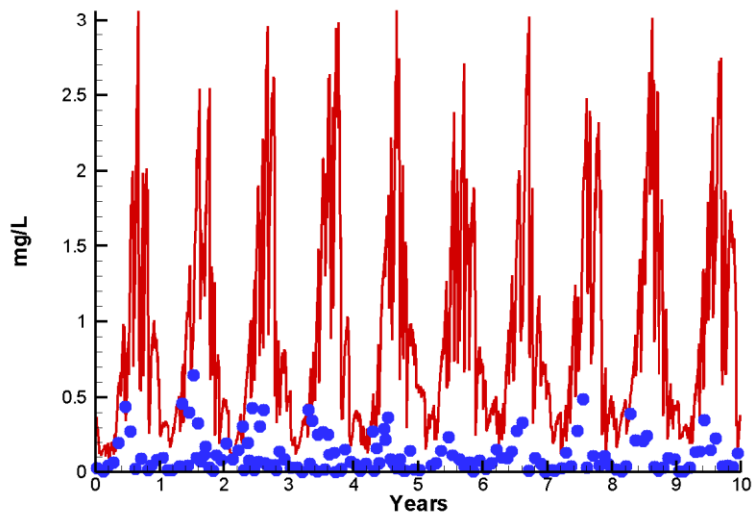
Calibration
Minimum vertical
diffusivity $0.014 \text{ cm}^2/\text{s}$

Sensitivity
Minimum vertical
diffusivity $0.22 \text{ cm}^2/\text{s}$

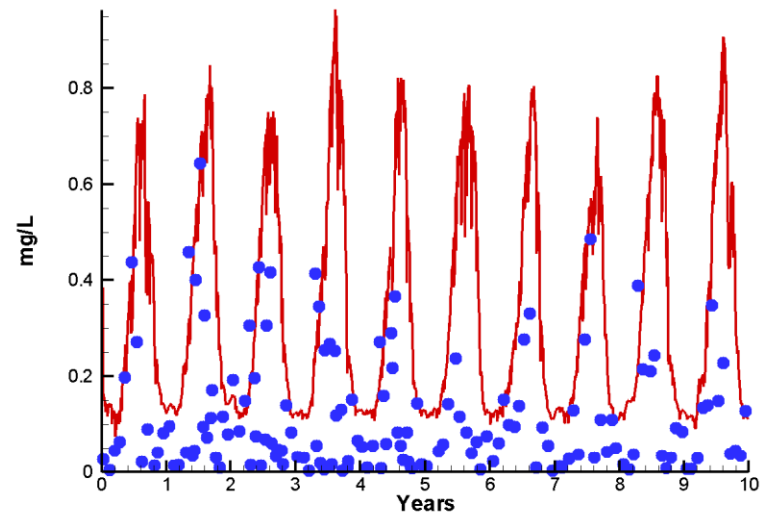
Run119 2002-2011
Dissolved Oxygen ET4.2 Bottom



Run117 2002-2011
Ammonium ET4.2 Bottom

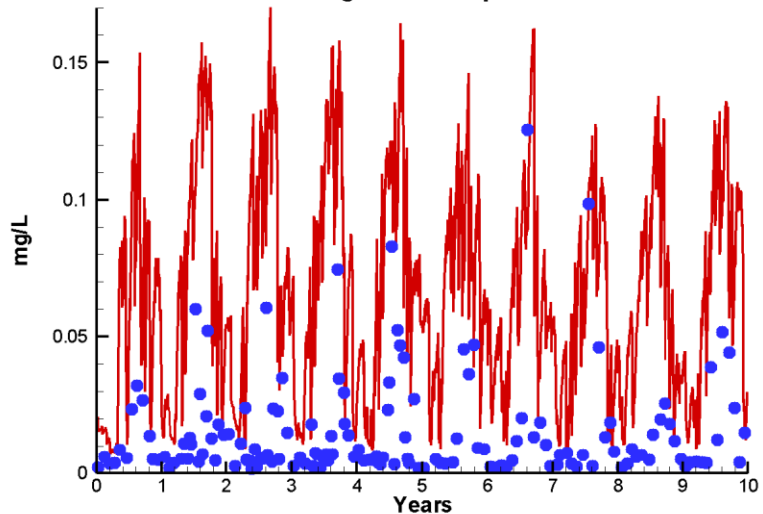


Run119 2002-2011
Ammonium ET4.2 Bottom



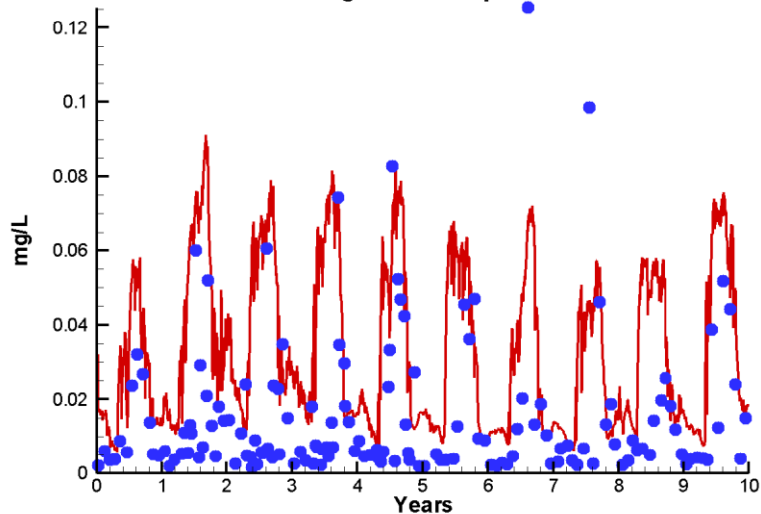
Calibration 0.014 cm²/s

Run117 2002-2011
Dissolved Inorganic Phosphorus ET4.2 Bottom

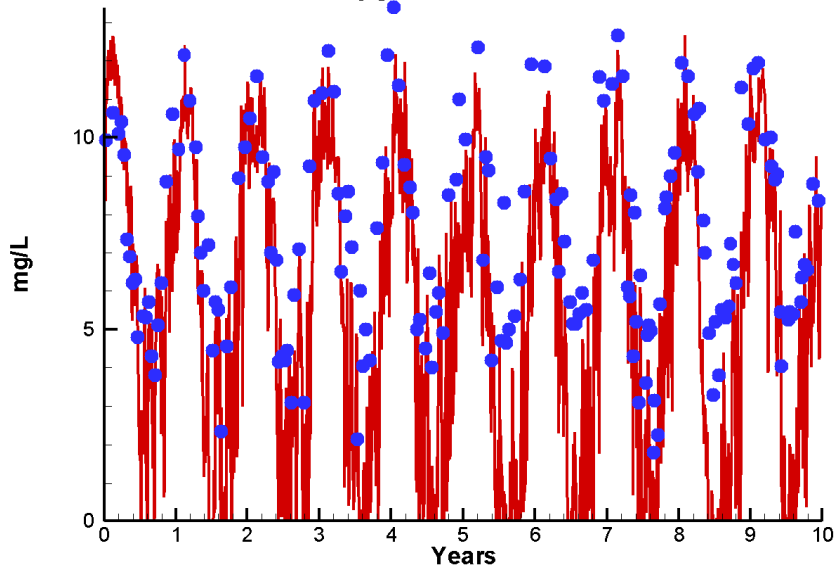


Sensitivity 0.22 cm²/s

Run119 2002-2011
Dissolved Inorganic Phosphorus ET4.2 Bottom



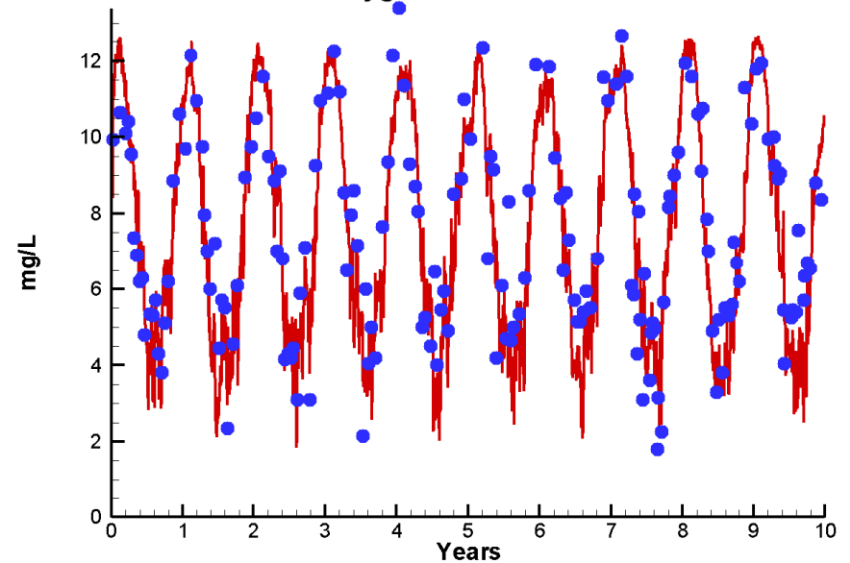
Run117 2002-2011
Dissolved Oxygen WT5.1 Bottom



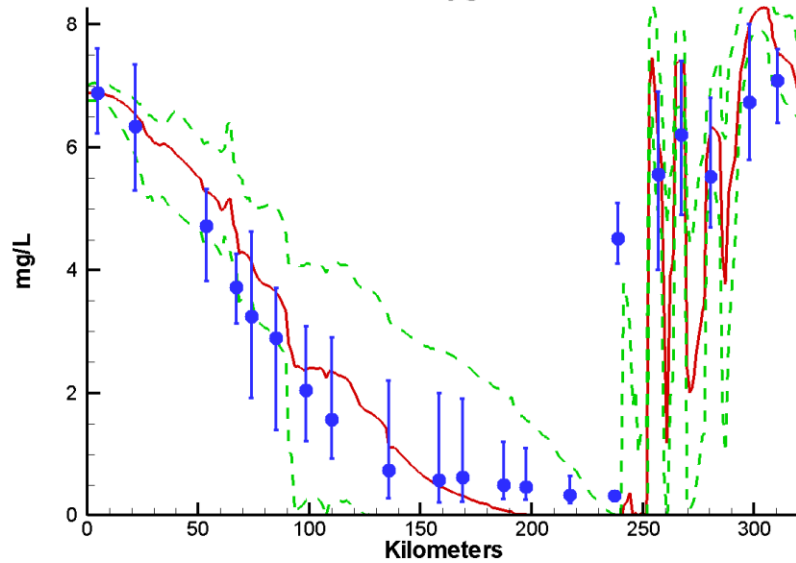
Calibration
Minimum vertical
diffusivity $0.014 \text{ cm}^2/\text{s}$

Sensitivity
Minimum vertical
diffusivity $0.22 \text{ cm}^2/\text{s}$

Run119 2002-2011
Dissolved Oxygen WT5.1 Bottom



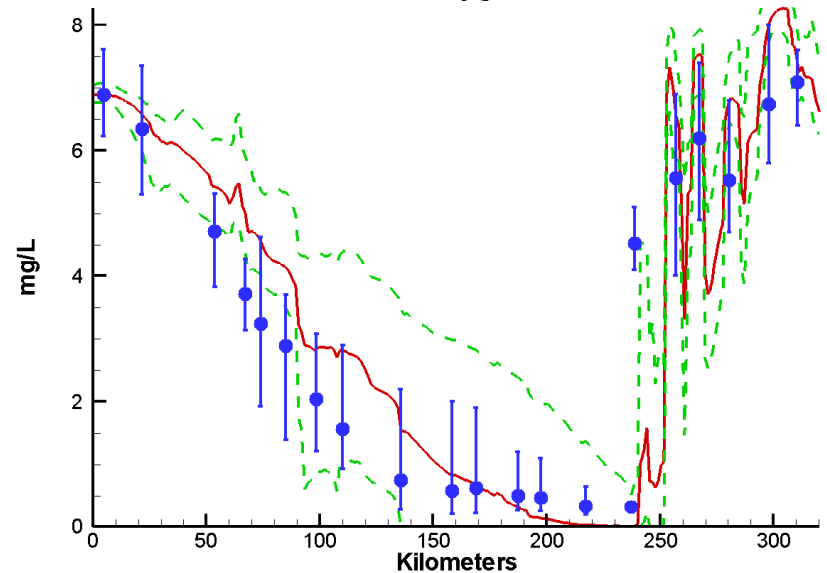
Mainstem Bay 2002-2011 Run117
Bottom Dissolved Oxygen Summer 2010



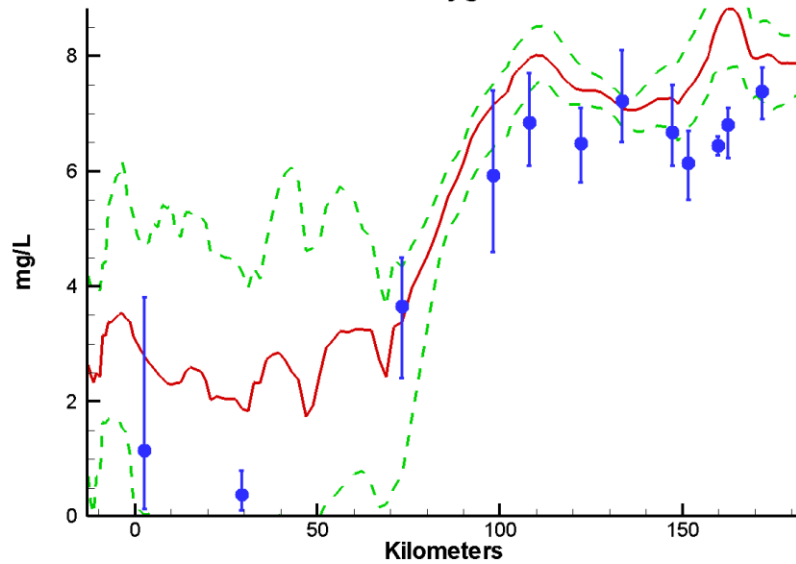
Calibration
Minimum vertical
diffusivity $0.014 \text{ cm}^2/\text{s}$

Sensitivity
Minimum vertical
diffusivity $0.22 \text{ cm}^2/\text{s}$

Mainstem Bay 2002-2011 Run119
Bottom Dissolved Oxygen Summer 2010



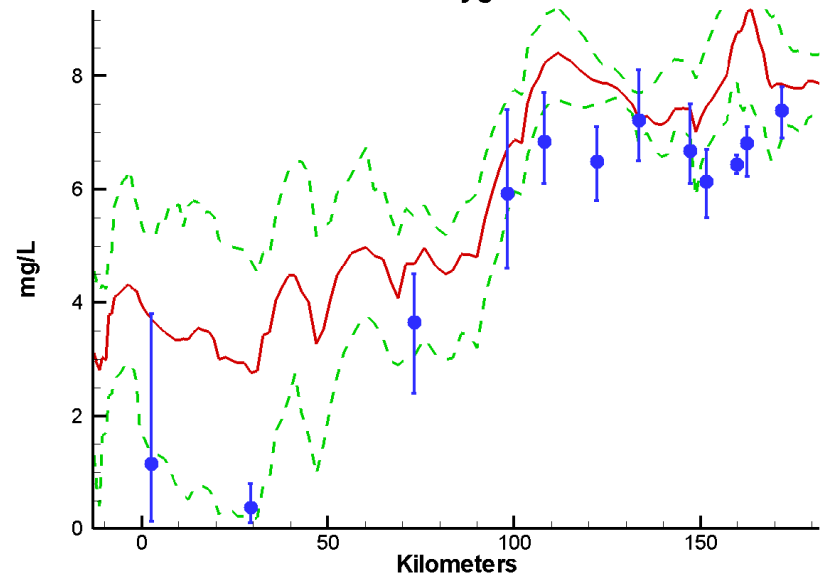
Potomac River 2002-2011 Run117
Bottom Dissolved Oxygen Summer 2007



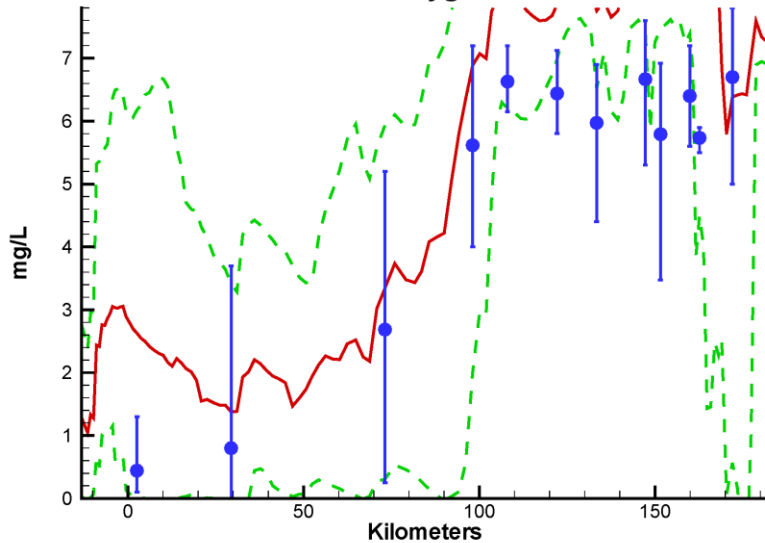
Calibration
Minimum vertical
diffusivity $0.014 \text{ cm}^2/\text{s}$

Sensitivity
Minimum vertical
diffusivity $0.22 \text{ cm}^2/\text{s}$

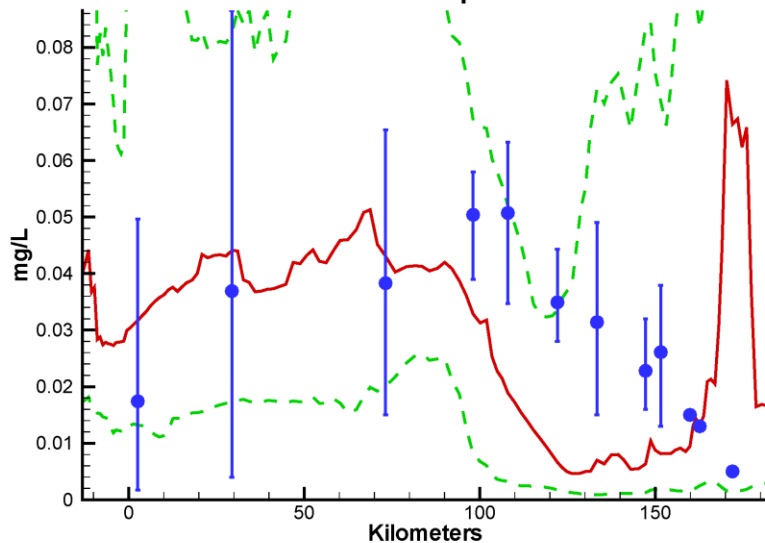
Potomac River 2002-2011 Run119
Bottom Dissolved Oxygen Summer 2007



Potomac River (Run 406)
Bottom Dissolved Oxygen Summer 1994



Potomac River (Run 406)
Bottom Dissolved Phosphate Summer 1994



We've known for a long time we needed to correct (MINVERT) vertical diffusivity in tidal fresh water. It looks like the need to adjust minimum vertical diffusivity is more widespread. There's a cost to doing this but it seems like the correct way forward.