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Resource managers need a practical perspective on **travel time** of groundwater to streams, detailed understanding of the sources of nitrogen for **targeting management** efforts and to better **quantify water-quality improvements**

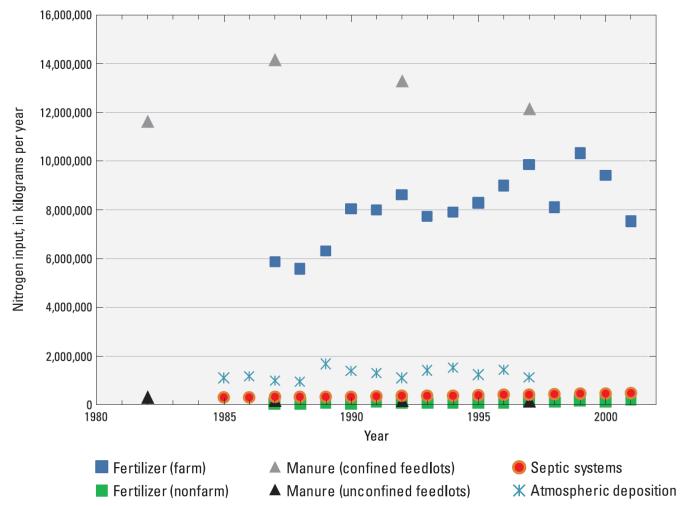


State of Delaware 2010 Combined Watershed Assessment Report (305(b)) and Determination for the Clean Water Act Section 303(d) List of Waters Needing TMDLs



Department of Natural Resources and Environmental Control April 1, 2010 Bucks Branch has some of the highest measured concentrations of total nitrogen in any stream in the State (Delaware Department of Natural Resources and Environmental Control, 2010) and most of the nitrogen is in the form of nitrate.

Introduction Sources



The vast majority of nitrogen inputs in this part of Sussex County, Delaware are from manure and fertilizer



Purpose/Objectives

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Concentrations of sulfur hexafluoride (SF₆), dissolved gases and silica in groundwater and surface water to determine the apparent age of groundwater in the aquifer and to estimate the average residence time of groundwater discharging to streams.



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chemical and isotopic analysis of nitrate

in the groundwater and surface water of the Bucks Branch watershed

Major-ion chemistry and the nitrogen and oxygen isotopic composition of nitrate in groundwater and stream water during base flow to evaluate the potential sources of nitrogen and to describe biogeochemical processes



Study Area Location



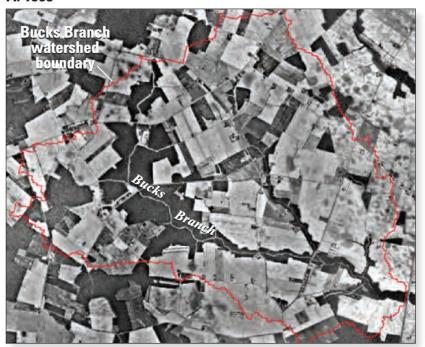
The study area is located approximately 32 miles (mi) west of the Atlantic Ocean shoreline in Sussex County, Delaware

Bucks Branch drains 7.02 square miles (mi²), and is a tributary to the Nanticoke River located on the Delmarva Peninsula, which eventually drains into the Chesapeake Bay.



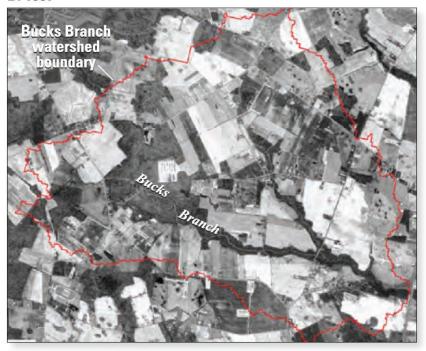
Study Area Historical

A. 1968



Aerial photograph by Agricultural Stabilization and Conservation Service, 1968

B. 1997

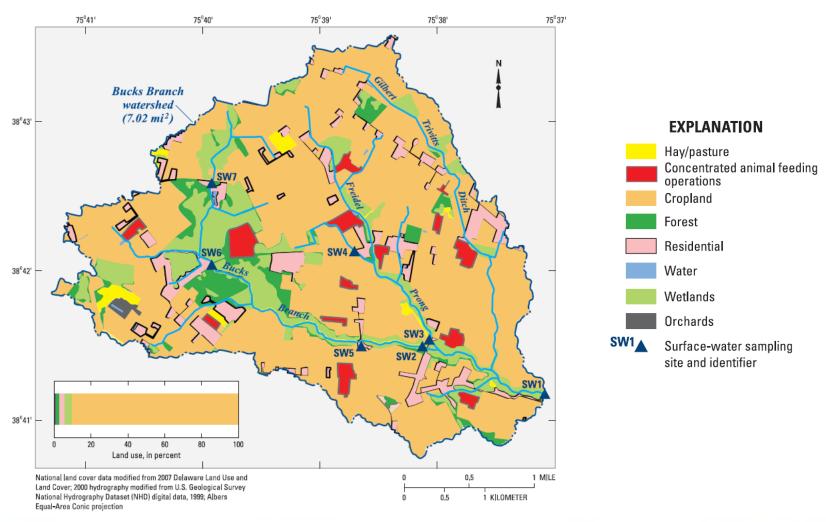


Aerial photograph by Earth Data (formerly PhotoScience), 1997



Study Area

Land Use





Morphology









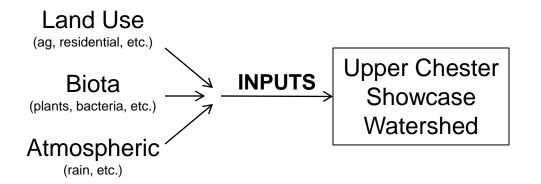
Channelization and dredging (ditching) of streams affects the nearby water table, impairs the functional properties of the aquatic ecosystem, reduces canopy cover

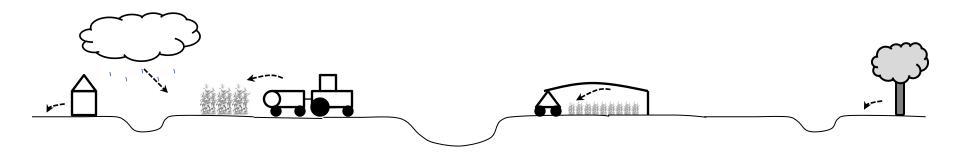




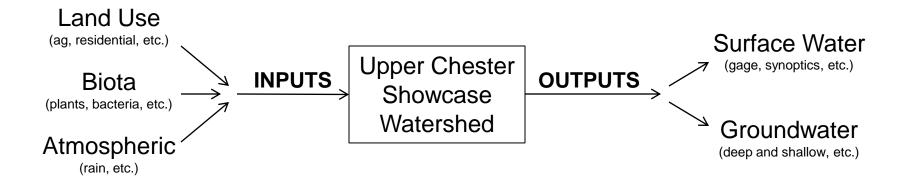


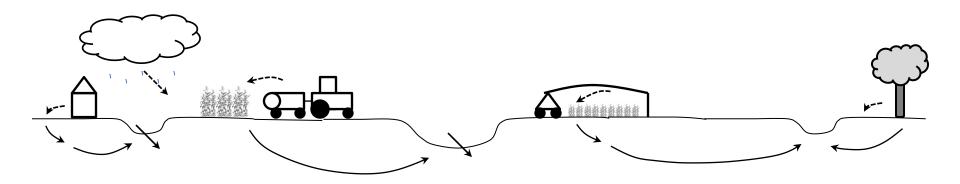
Mass Balance





Mass Balance





Methods Sampling



Stream Synoptic (n =7, 5)

High/Low Baseflow Conditions

Field Parameters

Major Ions (DNREC)

N & O Isotopes (USGS)

Groundwater Synoptic (n = 10)

Sampled Once

Field Parameters

Major Ions (DNREC)

N & O Isotopes (USGS)

Dissolved Gases (USGS)

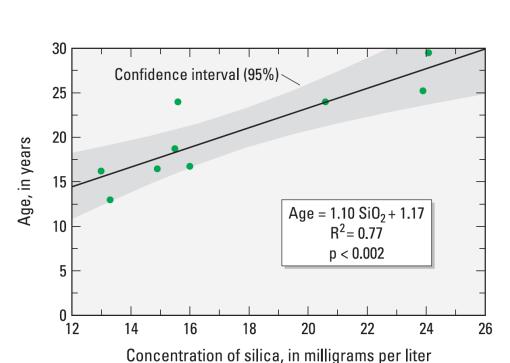


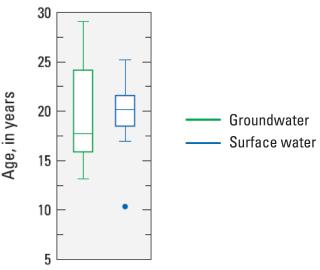




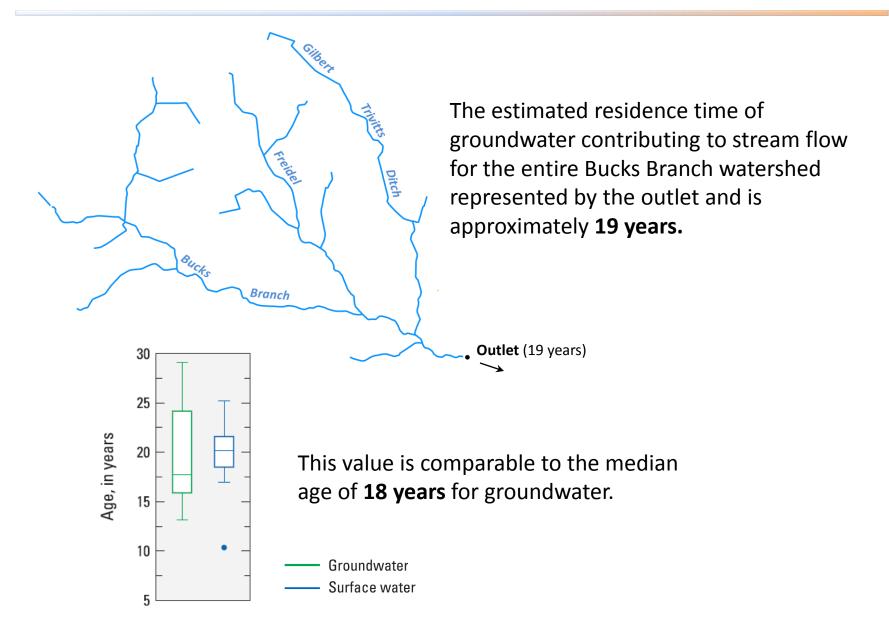
Residence Time of GW

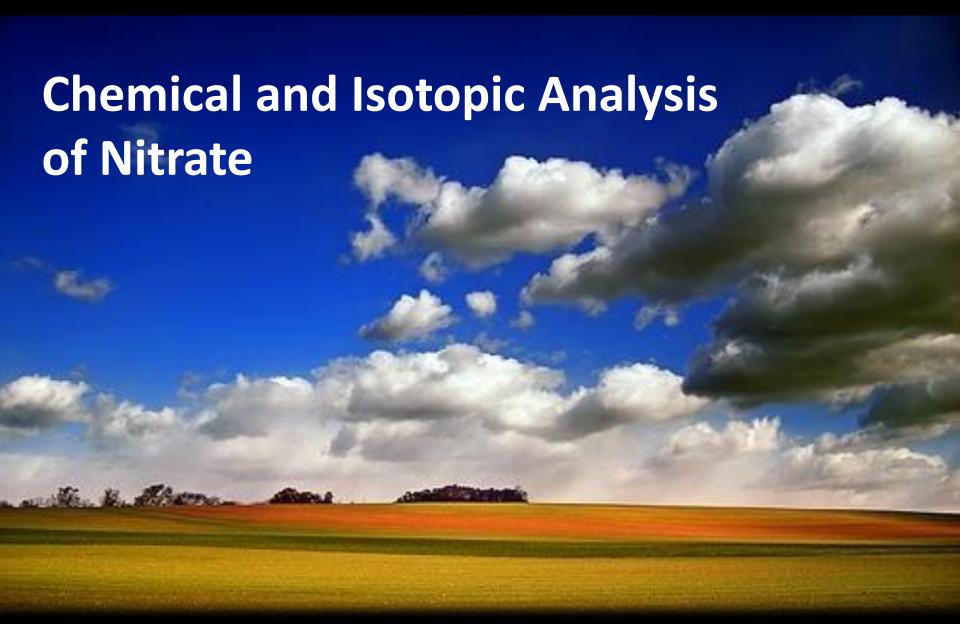
The apparent age of groundwater sampled ranged from 13 to 30 years, with a median age of **18 years**





The residence time of groundwater contributing to the **stream** during base flow was based on an empirical model using simple linear regression of **groundwater ages** and **silica** concentrations in groundwater

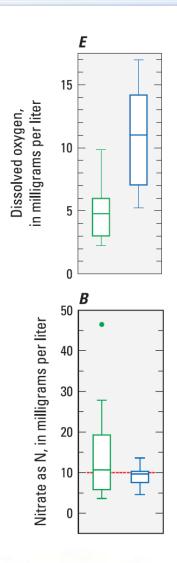






Major Ion Chemistry

Oxic Conditions



The surficial aquifer had **oxic conditions** with dissolved oxygen levels greater than 0.5 mg/L

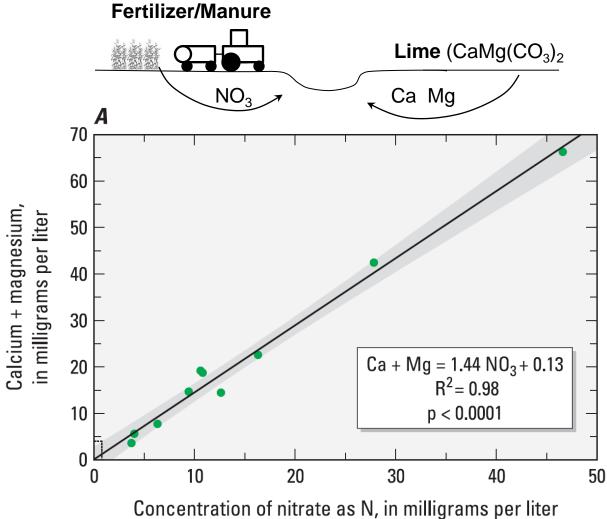
The lack of reducing conditions for effective removal of nitrate by microbes coupled with the high rate of nitrogen input often leaves any effects of possible loss of nitrate indistinguishable

Nitrate exceeded the USEPA drinkingwater standard (10 mg/L as N) in 60 percent of the groundwater wells and at 42 percent of the surface-water sites



Major Ion Chemistry

Applications



Fertilizer, manure and soil organic matter produce an acidity that is often neutralized with lime (crushed dolomite) [CaMg(CO₃)₂]

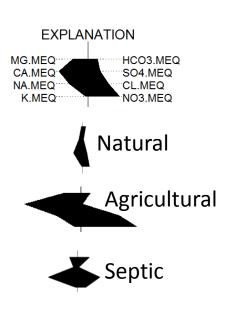
A similar relation of NO₃ was found with potash (KCI) and specific conductance.

EXPLANATION

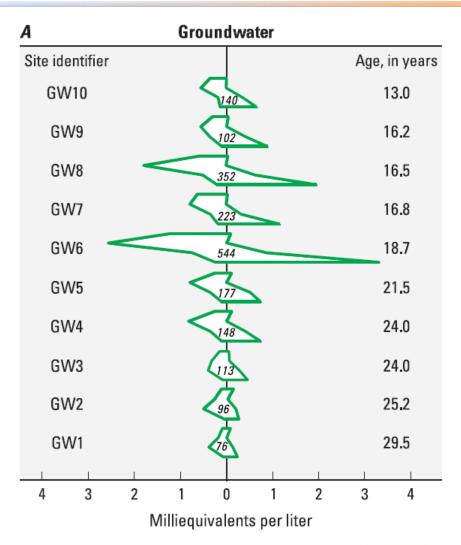
- Groundwater
- Natural conditions (Denver, 1986 and Denver, 1989)
- Confidence interval (95%)



Major Ion Chemistry

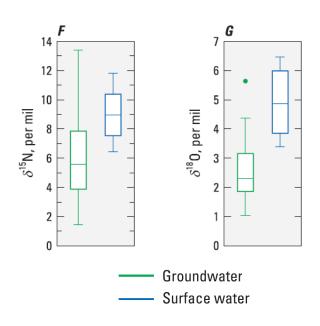


The relative abundance of cations and anions in ground water present a pattern commonly seen in water affected by agriculture

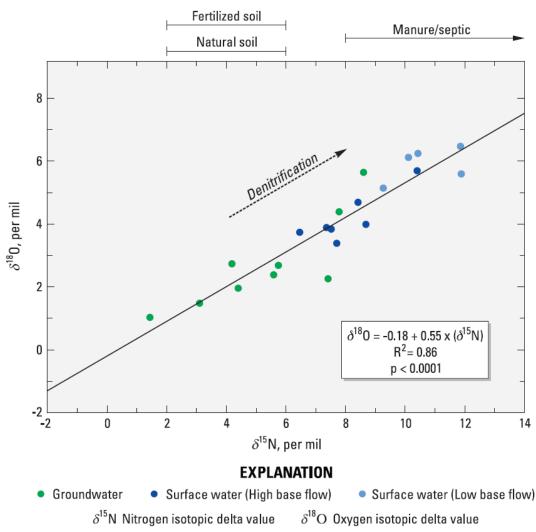




Isotopic Composition of Nitrate

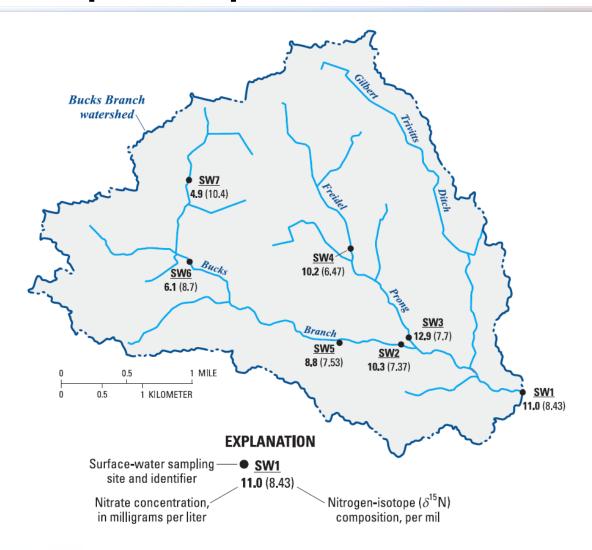


This linear relation between the N and O isotopes in flow path studies has been shown to be an indication of **denitrification** in groundwater





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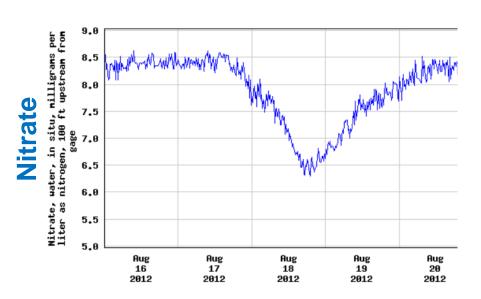


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- Geochemistry indicates that **agriculture** is **the dominant source** of **nitrate** contamination and the observed patterns are similar other studies
- Nitrogen and oxygen isotopes (δ^{15} N and δ^{18} O) of nitrate in groundwater and surface water indicate some **loss of nitrate through denitrification**, but this process is not sufficient to remove all of the nitrate and concentrations of nitrate remain elevated.

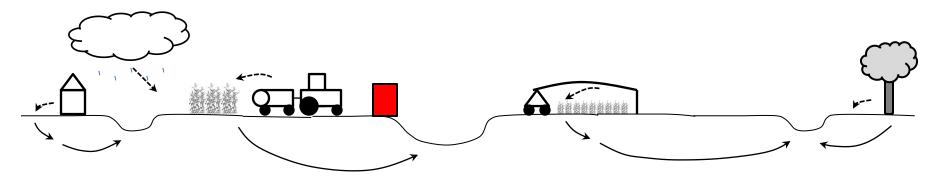


Gage Monitoring

- Monthly (Baseflow) Sampling
- Event (Storm) Sampling
- Major Ions, Nutrients
- N & O Isotopes
- Continuous WQ (Temp, pH, SC, DO)
- Nitrate Sensor



http://waterdata.usgs.go v/usa/nwis/uv?01493112

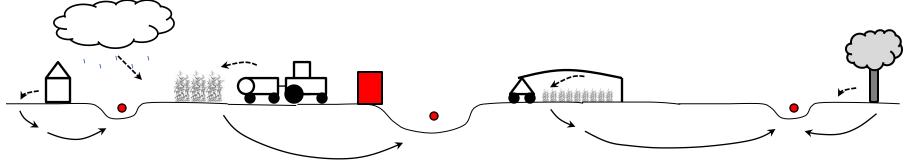


Possible Next Steps

Seasonal Synopotics

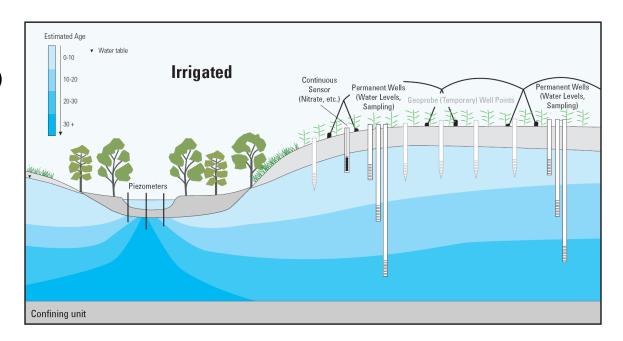
- High/Low Baseflow Conditions
- Field Parameters
- Major Ions, Nutrients
- N & O Isotopes
- Nitrate Sensor

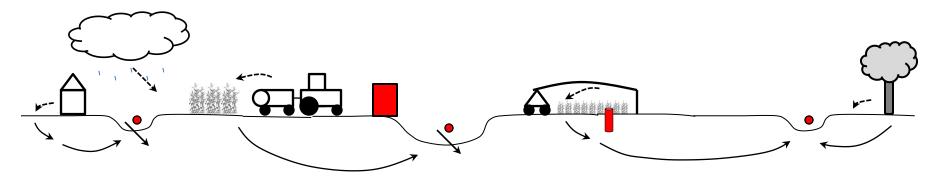




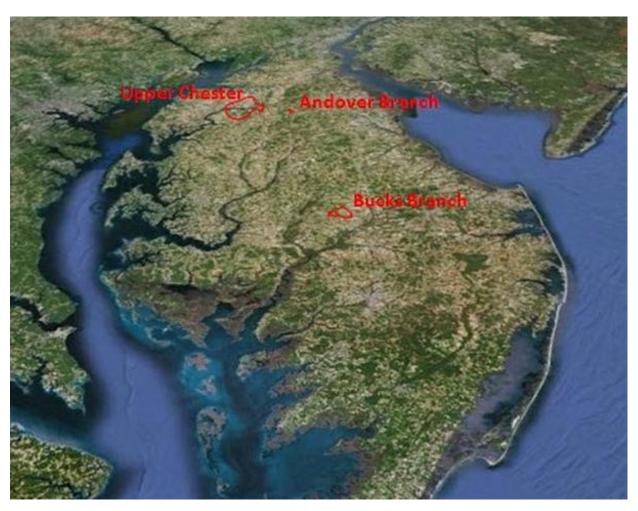
Possible Next Steps

- Bi-Monthly Sampling of Shallow Wells (WQ, Levels)
- Continuous Water Quality (Temp, SC, Nitrate)
- Geoprobe Synoptics
- Track Inputs





Monitoring the Water Quality Response of Conservation Practices in Small Watersheds



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The authors would like to thank Hassan Mirsajadi, Amber Joseph, Blair Venables, Xie Xia, Kathy A. Knowles, Sergio Huerta, and Ben Pressly of DNREC for overall project support.

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